



A Publication
of Reliable Methods
for the Preparation
of Organic Compounds

Working with Hazardous Chemicals

The procedures in *Organic Syntheses* are intended for use only by persons with proper training in experimental organic chemistry. All hazardous materials should be handled using the standard procedures for work with chemicals described in references such as "Prudent Practices in the Laboratory" (The National Academies Press, Washington, D.C., 2011; the full text can be accessed free of charge at http://www.nap.edu/catalog.php?record_id=12654). All chemical waste should be disposed of in accordance with local regulations. For general guidelines for the management of chemical waste, see Chapter 8 of Prudent Practices.

In some articles in *Organic Syntheses*, chemical-specific hazards are highlighted in red "Caution Notes" within a procedure. It is important to recognize that the absence of a caution note does not imply that no significant hazards are associated with the chemicals involved in that procedure. Prior to performing a reaction, a thorough risk assessment should be carried out that includes a review of the potential hazards associated with each chemical and experimental operation on the scale that is planned for the procedure. Guidelines for carrying out a risk assessment and for analyzing the hazards associated with chemicals can be found in Chapter 4 of Prudent Practices.

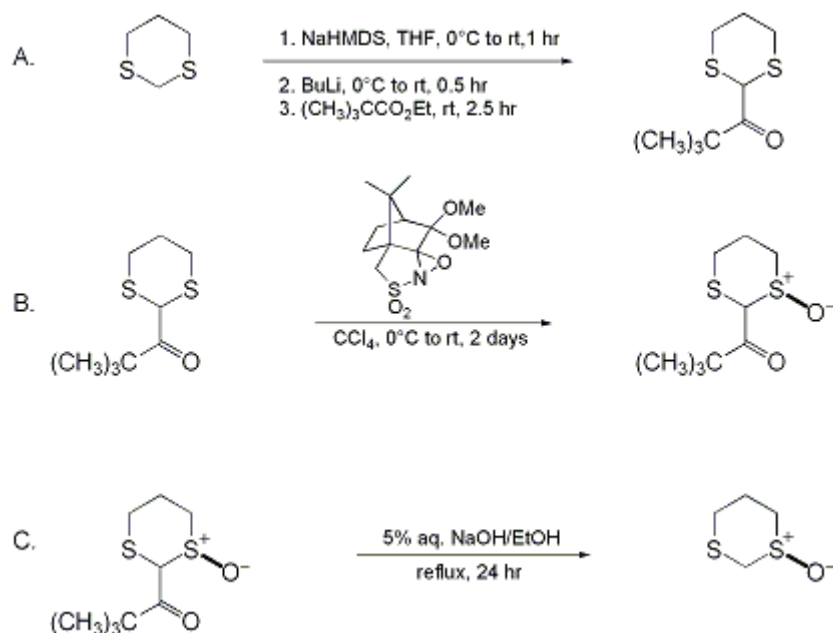
The procedures described in *Organic Syntheses* are provided as published and are conducted at one's own risk. *Organic Syntheses, Inc.*, its Editors, and its Board of Directors do not warrant or guarantee the safety of individuals using these procedures and hereby disclaim any liability for any injuries or damages claimed to have resulted from or related in any way to the procedures herein.

September 2014: The paragraphs above replace the section "Handling and Disposal of Hazardous Chemicals" in the originally published version of this article. The statements above do not supersede any specific hazard caution notes and safety instructions included in the procedure.

Organic Syntheses, Coll. Vol. 10, p.378 (2004); Vol. 76, p.37 (1999).

1S-(–)-1,3-DITHIANE 1-OXIDE

[1,3-Dithiane, 1-oxide, (S)-]



Submitted by Philip C. Bulman Page¹, Jag P. Heer¹, Donald Bethell¹, Eric W. Collington², and David M. Andrews².

Checked by William Moser and Amos B. Smith, III.

1. Procedure

A. 2-(2,2-Dimethylpropanoyl)-1,3-dithiane. To 43.0 g (0.358 mol) of 1,3-dithiane (Note 1) at 0°C under a nitrogen atmosphere is added 396 mL of a 1 M solution of sodium hexamethyldisilazide in tetrahydrofuran (THF) (Note 2) and (Note 3). The resulting yellow solution is allowed to reach room temperature and then stirred at room temperature for 1 hr. The solution is cooled to 0°C, and 172 mL of a 2.5 M solution of butyllithium in hexanes (Note 4) is added. The reaction mixture is allowed to warm to room temperature and then stirred at room temperature for 30 min. Ethyl 2,2-dimethylpropanoate (65.0 mL, 0.427 mol) (Note 5) is added, and the mixture is stirred at room temperature for 2.5 hr. An aqueous saturated solution of ammonium chloride (200 mL) is added, and the aqueous phase is extracted three times with 200 mL of dichloromethane (Note 6). The combined organic extracts are washed with 100 mL of water and dried over anhydrous magnesium sulfate. The solvents are removed under reduced pressure to give a yellow solid. Repeated trituration with petroleum ether 40–60°C (Note 7) followed by filtration gives 51.0–55.1 g (70–75%) of 2-(2,2-dimethylpropanoyl)-1,3-dithiane as colorless needles, mp 97–99°C (Note 8).

B. anti- and syn-1S-(2,2-Dimethylpropanoyl)-1,3-dithiane 1-oxide. (+)-[(8,8-Dimethoxycamphoryl)sulfonyl]oxaziridine (51.0 g, 0.176 mol) (Note 9) is added to a cooled, stirred solution of 36.0 g (0.176 mol) of 2-(2,2-dimethylpropanoyl)-1,3-dithiane in 1000 mL of carbon tetrachloride (Note 10) at 0°C. The reaction mixture is allowed to reach room temperature, and stirring is continued at room temperature for a further 48 hr. The reaction mixture is filtered to remove the bulk of the (+)-[(8,8-dimethoxycamphoryl)sulfonyl]imine, and the filtrate is evaporated to dryness under reduced pressure. The residue is purified by passage through a short column of silica gel using dichloromethane as initial eluant to remove residual (+)-[(8,8-dimethoxycamphoryl)sulfonyl]imine. The column is then flushed with ethyl acetate to give 29.8–33.5 g (77–86%) of an ca. 3:1 mixture of anti- and syn-1S-(2,2-dimethylpropanoyl)-1,3-dithiane 1-oxide as a colorless crystalline solid, mp 103–105°C (Note 11).

C. (1*S*)-(-)-1,3-Dithiane 1-oxide. A mixture of anti- and syn-1*S*-2-(2,2-dimethylpropanoyl)-1,3-dithiane 1-oxide (33 g, 0.150 mol) is dissolved in 500 mL of ethanol (Note 12), and 200 mL of aqueous 5% sodium hydroxide is added. The mixture is heated under reflux for 24 hr. The mixture is allowed to cool, and 500 mL of dichloromethane is added. The organic layer is separated, and the aqueous phase is extracted four times, with 100 mL of dichloromethane. The combined organic extracts are dried over anhydrous magnesium sulfate and evaporated to dryness under reduced pressure to give a beige solid. The solid is triturated with diethyl ether to give 13 g (64%) of 1*S*-(-)-1,3-dithiane 1-oxide as a colorless solid, mp 90-92°C (Note 13) and (Note 14).

2. Notes

- 1,3-Dithiane was stored in a desiccator over self-indicating silica gel.
- Tetrahydrofuran was distilled under nitrogen from the benzophenone ketyl radical.
- Sodium hexamethyldisilazide [sodium bis(trimethylsilyl)amide] was purchased from the Aldrich Chemical Company, Inc., in 100- or 800-mL bottles as a 1 M solution in tetrahydrofuran. Glassware used for moisture sensitive reactions was dried at 180°C and allowed to cool in a desiccator over self-indicating silica gel. Reactions were carried out under a slight positive static pressure of argon.
- Butyllithium was purchased from the Aldrich Chemical Company, Inc., in 800-mL bottles as a 2.5 M solution in hexanes; the molarity was determined by titration against a solution of diphenylacetic acid.
- Commercially available reagents were used as supplied unless otherwise stated.
- Dichloromethane was dried by distillation from calcium hydride.
- Petroleum ether (40-60°C) was distilled prior to use.
- The analytical data for 2-(2,2-dimethylpropanoyl)-1,3-dithiane are as follows: Found: C, 52.73; H, 7.87. C₉H₁₆OS₂ requires C, 52.90; H, 7.89%; IR (Nujol) cm⁻¹: 2900, 1673; ¹H NMR (400 MHz, CDCl₃) δ: 1.24 (s, 9 H), 1.95-2.09 (m, 1 H), 2.13-2.23 (m, 1 H), 2.56 (ddd, 2 H, J = 2.4, 7.0, 12.5), 3.43 (dt, 2 H, J = 2.4, 12.5), 4.51 (s, 1 H); m/z (EI) 204.06445 (M⁺); C₉H₁₆OS₂ requires 204.06425.
- For the preparation of (+)-[(8,8-dimethoxycamphoryl)sulfonyl]oxaziridine see: Chen, B.-C.; Murphy, C. K.; Kumar, A.; Reddy, R. T.; Clark, C.; Zhou, P.; Lewis, B. M.; Gala, D.; Mergelsberg, I.; Scherer, D.; Buckley, J.; DiBenedetto, D.; Davis, F. A. *Org. Synth., Coll. Vol. IX* **1998**, 212. A somewhat modified procedure³ is as follows: (+)-[(8,8-Dimethoxycamphoryl)sulfonyl]oxaziridine. Aliquat 336® (tri-*n*-octyl-methylammonium chloride) (5.0 mL, 10.9 mmol) is added to a stirred solution of 50.0 g (183 mmol) of (+)-[(8,8-dimethoxycamphoryl)sulfonyl]imine in 250 mL of dichloromethane at 0°C. A solution of 50.0 g (362 mmol) of potassium carbonate in 100 mL water is added and the biphasic reaction mixture is stirred for 5 min. A commercial solution (30% w/v) of hydrogen peroxide (83.0 mL, 732 mmol) is added dropwise over 30 min. The reaction is then allowed to warm to room temperature and stirred for about 6-7 hr (Note 15). The organic layer is separated and the aqueous phase extracted three times, each with 100 mL of dichloromethane. Residual hydrogen peroxide in the aqueous phase is carefully destroyed by the addition of saturated aqueous sodium sulfite. The combined organic extracts are rapidly washed with an aqueous solution of 5.0 g of sodium sulfite in 100 mL water and 100 mL of saturated brine and dried over anhydrous magnesium sulfate. Removal of the solvent under reduced pressure, at a bath temperature not exceeding 40°C, gives a white solid consisting of (+)-[(8,8-dimethoxycamphoryl)sulfonyl]oxaziridine contaminated with (+)-[(8,8-dimethoxycamphoryl)sulfonyl]imine. Recrystallization from absolute ethanol furnishes 51.3 g (97%) of (+)-[(8,8-dimethoxycamphoryl)sulfonyl]oxaziridine, mp 188-190°C (Note 16).
- Carbon tetrachloride was used as supplied without further purification.
- The analytical data for anti- and syn-1*S*-(2,2-dimethylpropanoyl)-1,3-dithiane 1-oxides are as follows: Found: C, 48.91; H, 7.35. C₉H₁₆O₂S₂ requires C, 49.06; H, 7.32; IR (Nujol) cm⁻¹: 2900, 1706, 1030; ¹H NMR (400 MHz, CDCl₃) δ for anti-: 1.26 (s, 9 H), 2.04-2.15 (m, 1 H), 2.45-2.70 (m, 2 H), 2.75-2.90 (m, 2 H), 3.44-3.56 (m, 1 H), 4.72 (s, 1 H); for syn-: 1.24 (s, 9 H), 2.22-2.35 (m, 1 H), 2.45-2.55 (m, 2 H), 3.00-3.15 (m, 2 H), 3.97 (dt, 1 H, J = 3.5, 13.8), 4.98 (s, 1 H); m/z (EI) 220.05931 (M⁺); C₉H₁₆OS₂ requires 220.059187; ee (anti) = 87%, ee (syn) = 88% from ¹H NMR studies (Note 17).
- Ethanol was used as supplied without further purification.
- The analytical data for 1*S*-(-)-1,3-dithiane 1-oxide are as follows: Found: C, 35.18; H, 5.93. C₄H₈OS₂ requires C, 35.27; H, 5.89; IR (Nujol) cm⁻¹: 2927, 1047; ¹H NMR (400 MHz, CDCl₃) δ: 2.10-2.35 (m, 1 H), 2.45-2.77 (m, 4 H), 3.35 (ddd, 1 H, J = 3.0, 6.0, 9.5), 3.66 (d, 1 H, J = 12.7), 4.03 (d, 1 H, J = 12.7); m/z (EI) 136.00151 (M⁺); C₄H₈OS₂ requires 136.00166; ee = 87% from ¹H NMR studies (Note 17).

14. The checkers obtained the product in about 54% yield and found flash chromatography to be more effective in its purification. This was accomplished using a 16-cm × 5-cm column of silica gel and CHCl₃/MeOH (96:4) as the eluant. With collection of ca. 50-mL fractions, the product was observed in fractions 12-21. Visualization of the product was accomplished by TLC (product R_f = 0.4 in CHCl₃/MeOH 96:4, [anisaldehyde](#) stain).

15. The checkers noted that complete oxidation typically required ca. 6-7 hr and recommend checking the progress of the reaction in the following way: a 1-mL aliquot is removed from the organic layer, diluted with 2 mL of [methylene chloride](#), and analyzed by TLC eluting with [methylene chloride](#) (I₂ visualization); imine R_f = 0.34, [oxaziridine](#) R_f = 0.51

16. The analytical data for (+)-[(8,8-dimethoxycamphoryl)sulfonyl]oxaziridine are as follows: Found: C, 49.77; H, 6.62; N, 4.88. C₁₂H₁₉NO₅S requires C, 49.83; H, 6.57; N, 4.84; IR (CH₂Cl₂ film) cm⁻¹: 1367, 1345, 1165; ¹H NMR (400 MHz, CDCl₃) δ: 1.06 (s, 3 H), 1.32 (s, 3 H), 1.75-2.30 (m, 5 H), 3.08 (d, 1 H, J = 12.0), 3.29 (d, 1 H, J = 12.0), 3.27 (s, 3 H), 3.34 (s, 3 H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ: 20.5, 21.6, 28.1, 29.3, 45.1, 47.4, 52.9, 50.5, 50.8, 54.6, 97.6, 102.8; m/z (CI) 290.10619 (MH⁺); C₁₂H₂₀NO₅S requires 290.10622; [α]_D²⁰ +91° (CHCl₃, c 3.00) ([Note 17](#)).

17. Optical rotations were measured on Optical Activity AA-1000 or polAAr 2001 polarimeters operating at 589 nm, corresponding to the [sodium D](#) line. Enantiomeric excesses were determined by ¹H NMR chiral shift reagent studies using 10 equiv of (R)-(-)- or (S)-(+)-2,2,2-trifluoro-1-(9-anthryl) ethanol (Pirkle reagent).

Waste Disposal Information

All toxic materials were disposed of in accordance with "Prudent Practices in the Laboratory"; National Academy Press; Washington, DC, 1995.

3. Discussion

Non-racemic chiral sulfoxides have become important as sources of chirality for asymmetric carbon-carbon bond formation.⁴ For example, we have developed 1,3-dithiane 1-oxide (DiTOX) units as effective moieties for stereocontrol of a range of carbonyl group reactions, including enolate alkylation and amination, Mannich reaction, reduction, and heterocycloaddition.⁵ While we have been able to prepare several 2-monosubstituted⁶ and 2,2-disubstituted-1,3-dithiane 1-oxides⁷ in high enantiomeric excesses (ee) on scales of a few grams, we had difficulty until recently in preparing the parent compound, [1,3-dithiane 1-oxide](#), with very high ee in quantities of more than ca. 5 g.⁸ Enantiomerically pure [1,3-dithiane 1-oxide](#) has previously been prepared via adducts with (+)-camphor,⁹ and, by ourselves, using modified Sharpless oxidation techniques.^{8,10,11}

We have recently reported that [(8,8-dimethoxycamphoryl)sulfonyl]oxaziridine is a particularly effective reagent for asymmetric sulfide oxidation, especially in non-aryl sulfide substrates.³ Here we report a three-step chemical synthesis of [1,3-dithiane 1-oxide](#) with very high ee that is based upon such an oxidation as the key step. The procedure is effective for production of multigram quantities of material of either absolute configuration. The sequence is illustrated for the preparation of [1S-\(-\)-1,3-dithiane 1-oxide](#).

The route is based upon an acylation-oxidation-deacylation sequence, with commercially available, inexpensive [1,3-dithiane](#) employed as the starting material. 2-Acyl-1,3-dithianes have proved to be particularly effective substrates for asymmetric oxidation in our hands,^{8,10,3,12} and as [2-\(2,2-dimethylpropanoyl\)-1,3-dithiane](#) undergoes this asymmetric oxidation most efficiently (ca. 90% ee), it was chosen as the intermediate.

References and Notes

1. Robert Robinson Laboratories, Department of Chemistry, University of Liverpool, Oxford Street, Liverpool L69 3BX, England. This investigation has enjoyed the support of the EPSRC and Glaxo Research & Development (CASE award to JPH); present address: Department of Chemistry, Loughborough University, Loughborough, Leicestershire LE11 3TU, England

2. Glaxo Research & Development, Gunnels Wood Road, Stevenage, Hertfordshire SG13 9NJ, England.
3. Page, P. C. B.; Heer, J. P.; Bethell, D.; Collington, E. W.; Andrews, D. M. *Tetrahedron: Asymmetry* **1995**, 6, 2911. For a convenient procedure for the preparation of camphorsulfonyloxaziridines see: Page, P. C. B.; Heer, J. P.; Bethell, D.; Lund, A.; Collington, E. W.; Andrews, D. M. *J. Org. Chem.* **1997**, 62, 6093-6094.
4. Solladié, G. *Synthesis*, **1981**, 185; Posner, G. H. In "Asymmetric Synthesis"; Morrison, J. D., Ed.; Academic Press: New York, 1983; Vol. 2; Barbachyn, M. R.; Johnson, C. R. In "Asymmetric Synthesis"; Morrison, J. D.; Scott, J. W., Eds.; Academic Press: New York, 1983; Vol. 4; Nudelman, A. In "The Chemistry of Optically Active Sulfur Compounds", Gordon and Breach: New York, 1984; Posner, G. H. In "The Chemistry of Sulphones and Sulfoxides"; Patai, S.; Rappoport, Z.; Stirling, C., Eds.; Wiley, 1988; p. 823; Andersen, K. K. In "The Chemistry of Sulphones and Sulfoxides", Patai, S.; Rappoport, Z.; Stirling, C., Eds.; Wiley, 1988; p. 55; Kagan, H. B.; Rebiere, F. *Synlett* **1990**, 643; Aggarwal, V. K.; Thomas, A.; Franklin, R. J. *J. Chem. Soc., Chem. Commun.* **1994**, 1653; Solladié, G.; Carreño, M. E. In "Organosulfur Chemistry: Synthetic Aspects"; Page, P., Ed.; Academic Press: London, 1995; Vol. 1.
5. Page, P. C. B.; McKenzie, M. J.; Buckle, D. R. *J. Chem. Soc., Perkin Trans. I* **1995**, 2673, and references therein.
6. Page, P. C.; Namwindwa, E. S.; Klair, S. S.; Westwood, D. *Synlett* **1990**, 457; Page, P. C. B.; Wilkes, R. D.; Barkley, J. V.; Witty, M. J. *Synlett* **1994**, 547.
7. Page, P. C. B.; Wilkes, R. D.; Namwindwa, E. S.; Witty, M. J. *Tetrahedron* **1995**, 51, 2125.
8. Page, P. C. B.; Wilkes, R. D.; Witty, M. J. *Org. Prep. Proced. Int.* **1994**, 26, 702.
9. Bryan, R. F.; Carey, F. A.; Dailey, Jr., O. D.; Maher, R. J.; Miller, R. W. *J. Org. Chem.* **1978**, 43, 90.
10. Page, P. C. B.; Gareh, M. T.; Porter, R. A. *Tetrahedron: Asymmetry* **1993**, 4, 2139.
11. Kagan, H. B.; Rebiere, F. *Synlett* **1990**, 643, and references therein.
12. Page, P. C. B.; Heer, J. P.; Bethell, D.; Collington, E. W.; Andrews, D. M. *Synlett* **1995**, 773.

Appendix

Chemical Abstracts Nomenclature (Collective Index Number); (Registry Number)

1S(-)-1,3-Dithiane 1-oxide:
1,3-Dithiane, 1-oxide, (S)- (10); (63865-78-1)

2-(2,2-Dimethylpropanoyl)-1,3-dithiane:
1-Propanone, 1-(1,3-dithian-2-yl)-2,2-dimethyl- (10); (73119-31-0)

1,3-Dithiane:
m-Dithiane (8);
1,3-Dithiane (9); (505-23-7)

Sodium hexamethyldisilazide (NHMDs): Aldrich:
Sodium bis(trimethylsilyl)amide:
Disilazane, 1,1,1,3,3,3-hexamethyl-, sodium salt (8);
Silanamine, 1,1,1-trimethyl-N-(trimethylsilyl)-, sodium salt (9); (1070-89-9)

Butyllithium:
Lithium, butyl- (8,9); (109-72-8)

Ethyl 2,2-dimethylpropanoate: Aldrich: See:
Ethyl trimethylacetate:

Propanoic acid, 2,2-dimethyl-, ethyl ester (9); (3938-95-2)

anti-1S-(2,2-Dimethylpropanoyl)-1,3-dithiane 1-oxide:
1-Propanone, 2,2-dimethyl-1-(1-oxido-1,3-dithian-2-yl)-, (1S-trans)- (13); (160496-17-3)

(+)-[(8,8-Dimethoxycamphoryl)sulfonyl]oxaziridine:
4H-4a,7-Methanooxazirino[3,2-i][2,1]benzothiazole, tetrahydro-8,8-dimethoxy-9,9-dimethyl-, 3,3-dioxide, [2R-(2 α ,4 $\alpha\alpha$,7 α , 8aR)]- (12); (131863-82-6)

Aliquat 336: Methyltri-n-octylammonium chloride:
Ammonium, methyltrioctyl-, chloride (8);
1-Octanaminium, N-methyl-N,N-dioctyl-, chloride (9); (5137-55-3)

Hydrogen peroxide (8,9); (7722-84-1)