

A Publication of Reliable Methods for the Preparation of Organic Compounds

# **Working with Hazardous Chemicals**

The procedures in Organic Syntheses are intended for use only by persons with proper training in experimental organic chemistry. All hazardous materials should be handled using the standard procedures for work with chemicals described in references such as "Prudent Practices in the Laboratory" (The National Academies Press, Washington, D.C., 2011; the full accessed text can be free of charge at http://www.nap.edu/catalog.php?record\_id=12654). All chemical waste should be disposed of in accordance with local regulations. For general guidelines for the management of chemical waste, see Chapter 8 of Prudent Practices.

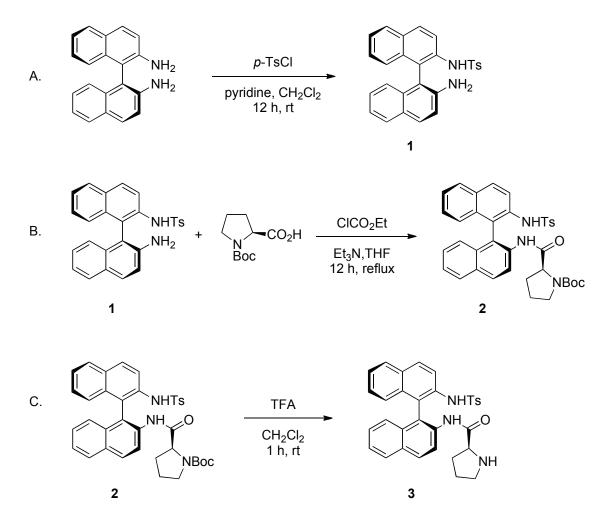
In some articles in *Organic Syntheses*, chemical-specific hazards are highlighted in red "Caution Notes" within a procedure. It is important to recognize that the absence of a caution note does not imply that no significant hazards are associated with the chemicals involved in that procedure. Prior to performing a reaction, a thorough risk assessment should be carried out that includes a review of the potential hazards associated with each chemical and experimental operation on the scale that is planned for the procedure. Guidelines for carrying out a risk assessment and for analyzing the hazards associated with chemicals can be found in Chapter 4 of Prudent Practices.

The procedures described in *Organic Syntheses* are provided as published and are conducted at one's own risk. *Organic Syntheses, Inc.,* its Editors, and its Board of Directors do not warrant or guarantee the safety of individuals using these procedures and hereby disclaim any liability for any injuries or damages claimed to have resulted from or related in any way to the procedures herein.

September 2014: The paragraphs above replace the section "Handling and Disposal of Hazardous Chemicals" in the originally published version of this article. The statements above do not supersede any specific hazard caution notes and safety instructions included in the procedure.

Copyright © 2011 Organic Syntheses, Inc. All Rights Reserved

## (*S*<sub>a</sub>,*S*)-*N*-[2'-(4-METHYLPHENYLSULFONAMIDO)-1,1'-BINAPHTHYL-2-YL]PYRROLIDINE-2-CARBOXAMIDE: AN ORGANOCATALYST FOR THE DIRECT ALDOL REACTION



Submitted by Santiago F. Viózquez,<sup>1</sup> Gabriela Guillena,<sup>1</sup> Carmen Nájera,<sup>1</sup>Ben Bradshaw,<sup>2</sup> Gorka Etxebarria-Jardi,<sup>2</sup> and Josep Bonjoch.<sup>2</sup> Checked by David Hughes.<sup>3</sup>

## 1. Procedure

A.  $(S_a)$ -N-[2'-Amino-(1,1'-binaphthyl)-2-yl]-4-methylbenzenesulfonamide (1). A 250-mL round-bottomed flask equipped with a 3-cm oval PTFE-coated magnetic stir bar is charged with  $(S_a)$ -(-)-1,1'-binaphthyl-2,2'diamine (3.13 g, 11.0 mmol, 1.0 equiv), dichloromethane (130 mL), and pyridine (10 mL, 124 mmol, 11 equiv). To the stirred solution is added *p*toluenesulfonyl chloride (2.03 g, 10.7 mmol, 0.97 equiv) in one portion (Note 1). The flask is sealed with a rubber septum through which is inserted *Org. Synth.* **2011**, *88*, 317-329 Published on the Web 4/4/2011 an 18-gauge inlet needle, which is connected to a nitrogen line and a gas bubbler, and a thermocouple probe (Note 2). The brown solution is stirred at 22 °C for 10 h (Note 3). The reaction solution is concentrated by rotary evaporation (40 °C bath temperature, 20 mmHg) to an oil that is transferred to a 500-mL separatory funnel with EtOAc (200 mL). The organic layer is washed with 2M HCl ( $5\times30$  mL) (Note 4), then vacuum-filtered through a bed of sodium sulfate (40 g) in a 150-mL medium-porosity sintered glass funnel. The filter cake is washed with EtOAc (2 x 40 mL). The filtrate is concentrated in a 500-mL round-bottomed flask by rotary evaporation (40 °C bath, 20 mmHg), then further dried under vacuum (20 mmHg) at room temperature for 14 h to afford 1 as a pink foam (4.7 g, 82% purity, 80% yield) which is used directly in the next step (Notes 5 and 6).

 $(S_a, S)$ -t-Butyl 2-[(2'-(4-methylphenylsulfonamido)-(1,1'-В. (2). *binaphthyl*)-2-*y*l-*carbamoyl*]*pyrrolidine-*1-*carboxylate* А 250-mL round-bottomed flask equipped with a 3-cm PTFE-coated magnetic stir bar is charged with (S)-N-(t-butoxycarbonyl)-L-proline (3.00 g, 13.9 mmol, 1.6 equiv), anhydrous THF (100 mL), and triethylamine (1.42 g, 14.0 mmol, 1.6 equiv) (Note 7). The flask is sealed with a rubber septum through which is inserted an 18-gauge inlet needle, which is connected to a nitrogen line and a gas bubbler, and a thermocouple probe (Note 2). The mixture is cooled to 3 °C with an ice-water bath and ethyl chloroformate (1.43 g, 13.2 mmol, 1.5 equiv) is added dropwise via a 3-mL syringe over 3 min where upon a fine white precipitate is formed (Note 8). The suspension is stirred 30 min at 0-5 °C and then a solution of 1 (4.7 g, 82 wt%, 3.85 assay g, 88 mmol, 1.0 equiv) in anhydrous THF (25 mL) is added dropwise over 5 min via a 40mL syringe. After the addition, the rubber septum is replaced with a condenser fitted with a gas adapter connected to a nitrogen line and gas bubbler. The mixture is refluxed using a heating mantle for 12 h (Note 9). At the end of the reaction, the suspension is cooled to room temperature, filtered through a 60-mL medium-porosity sintered glass funnel, and the filter cake is washed with THF (2×25 mL). The combined filtrates are concentrated by rotary evaporation (40 °C bath temperature, 20 mmHg) in a 500-mL round bottomed flask and further dried under vacuum (20 mmHg) for 3 h to provide 2 (7.7 g, estimated 65% purity, 5.0 assay g, 90% yield) as a pink foam which is used directly in the next step (Notes 10-12).

C.  $(S_a,S)$ -N-[2'-(4-Methylphenylsulfonamido)-1,1'-binaphthyl-2-ylpyrrolidine-2-carboxamide (**3**). The 500-mL flask containing crude **2** (7.7 g, 65% purity, 7.9 mmol) from the previous step is equipped with a 3-cm oval

PTFE-coated magnetic stir bar and charged with dichloromethane (80 mL) (Note 13). The flask is sealed with a rubber septum through which is inserted an 18-gauge inlet needle, which is connected to a nitrogen line and a gas bubbler, and a thermocouple probe (Note 2). Trifluoroacetic acid (16 mL, 208 mmol, 26 equiv) is added dropwise via a 20-mL syringe over 3 min, and the mixture is stirred at 20-22 °C for 1 h (Notes 14 and 15). At the end of reaction, the solution is cooled to 3 °C using an ice-water bath and the septum is removed and replaced with a 125-mL addition funnel to which is added 2.5 M sodium hydroxide (80 mL). The NaOH solution is added dropwise to the reaction mixture over 10 min (Notes 16 and 17). The mixture is transferred to a 250-mL separatory funnel, and the bottom organic layer is separated. The aq. layer is back extracted with dichloromethane (40 mL). The organic layers are combined and dried with sodium sulfate (100 g) (Note 18), then vacuum filtered through a 150-mL medium porosity sintered glass funnel into a 500-mL round bottomed flask. The filter cake is washed with dichloromethane (2x60 mL). The combined filtrate is concentrated by rotary evaporation (40 °C bath, 20 mmHg) to  $\sim$ 80 mL, then silica gel (30 g) is added, and the mixture is evaporated to a free-flowing powder. The material is purified by column chromatography (Note 19) with a final concentration in a 250-mL round bottom flask. The material is dried under vacuum (room temperature, 20 mmHg, 14 h) to provide 3 as a white solid (3.9–4.1 g). A 3-cm oval PTFE-coated stir bar and dichloromethane (8 mL) are added to the flask and the contents are warmed in a 40 °C water bath to dissolve the product, and then the flask is cooled to room temperature and is equipped with a 60-mL addition funnel. The mixture is stirred as hexanes (40 mL) are added through the addition funnel over 30 min. Crystallization occurs after 10 mL of hexanes is added, and the mixture becomes a thick slurry as the remainder of the hexanes is added. The slurry is stirred 12 h at ambient temperature and then is vacuum-filtered into a 60-mL sintered glass funnel. The filter cake is washed with 5:1 hexanes: dichloromethane (15 mL) and then is air-dried to constant weight to afford  $(S_a,S)-N-[2'-(4$ methylphenylsulfonamido)-1,1'-binaphthyl-2-yl-pyrrolidine-2-carboxamide (3) as a white crystalline solid (3.8–4.0 g, step yield 89–94%, 3-step yield 65-68%) (Notes 20-22).

1. The following reagents and solvents in Step A were used as received:  $(S_a)$ -(-)-1,1'-binaphthyl-2,2'-diamine (Sigma-Aldrich, 99%), dichloromethane (Fisher, ACS reagent, 99.5%), EtOAc (Fisher, ACS reagent, 99%), pyridine (Sigma-Aldrich, 99%), and *p*-TsCl (Acros, 99%).

2. The internal temperature was monitored using a J-Kem Gemini digital thermometer with a Teflon-coated T-Type thermocouple probe (12-inch length, 1/8 inch outer diameter, temperature range -200 to +250 °C). There was no exotherm on addition of *p*-TsCl.

3. The reaction was monitored by thin layer chromatography on silica gel (EMD, silica gel, grade 60,  $F_{254}$ ) with 1:1 EtOAc:hexanes as the eluent and visualization with UV. The diamine starting material has  $R_f = 0.5$  (blue fluorescence) and the tosyl product has  $R_f = 0.6$ . The bis-tosyl by-product co-elutes with the mono-tosylate product. The mono- and bis-tosyl products can be separated by TLC by using an eluent of 1:6 EtOAc:hexanes and 3 elutions (bis-Ts,  $R_f = 0.45$ ; mono-Ts,  $R_f = 0.40$ ).

4. The acid wash removes unreacted  $(S_a)$ -(-)-1,1'-binaphthyl-2,2'diamine, which can be recovered as follows. The combined acidic washes are neutralized with 2.5N NaOH until pH 8-10 and then are extracted with dichloromethane (3 × 30 mL). The combined organic layers are washed with brine, dried over sodium sulfate and concentrated by rotary evaporation (40 °C, 20 mmHg) to give 0.30 g (10%) of ( $S_a$ )-Binam.

The mono-tosylate 1 is approximately 82% pure (80% yield), 5. containing 7 wt% EtOAc and 11 wt% of the bis-tosylate by-product as determined by <sup>1</sup>H NMR analysis of the Ts methyl groups of the mono- and bis-tosylated species: <sup>1</sup>H NMR (400 MHz,  $CDCl_3$ ); mono-tosylate 1:  $\delta$ : 2.32; bis-tosylate:  $\delta$ : 2.40. This bis-tosylate by-product is difficult to separate by standard column chromatography due to overlap of the peaks using a more polar eluent and tailing of the early eluting bis-tosylate into the monotosylate peak using a less polar eluent. The submitters reported preparation of a purified sample by column chromatography on silica gel eluting with hexane/EtOAc, 6:1, following a literature report.<sup>4</sup> The checker purified a 100 mg crude sample by reverse-phase preparative HPLC using the following conditions: column, YMC-pack ODS-AQ, 5um, 150×20 mm I.D; mobile phase, linear gradient elution: 25% MeCN/75% water to 55% MeCN/45% water over 15 min; flow rate, 25mL/min; sample dissolved in MeCN at 10mg/mL; 1 mL per injection. Fractions eluting between 7-9 min were concentrated by rotary evaporation to remove the organic phase (bath temperature 35 °C, 10 mmHg). The remaining aqueous layer was lyophilized, affording 24 mg of mono-tosylate **1**.

6. Mono-tosylate **1** has the following spectroscopic properties: <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$ : 2.32 (s, 3 H, CH<sub>3</sub>), 3.29 (br s, 2 H, NH<sub>2</sub>), 6.42 (d, *J* = 8.4 Hz, 1 H), 6.67 (s, 1 H, NH), 6.94–7.09 (m, 5 H), 7.20–7.25 (m, 2 H), 7.37–7.43 (m, 3 H), 7.78 (d, *J* = 8.0 Hz, 1 H), 7.84 (t, *J* = 8.7 Hz, 1 H), 7.87 (d, *J* = 8.0 Hz, 1 H), 7.96 (d, *J* = 9.0 Hz, 1 H), 8.13 (d, *J* = 9.0 Hz, 1 H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$ : 21.7, 109.9, 118.2, 119.7, 121.8, 122.7, 123.6, 125.6, 125.9, 127.35, 127.37, 127.5, 128.37, 128.39, 129.7, 129.9, 130.9, 131.5, 133.0, 133.8, 133.9, 136.5, 142.9, 143.8; LC-MS calcd for [M + H]<sup>+</sup> 439.5; found, 439.2.

7. The following reagents and solvents in Step B were purchased from Sigma-Aldrich and used as received: N-(*t*-butoxycarbonyl)-*L*-proline (>99%), triethylamine (99.5%), THF (ACS reagent, >99%, inhibited with 250 ppm BHT), and ethyl chloroformate (97%).

8. Addition of ethyl chloroformate results in a slight exotherm from 3  $^{\circ}$ C to 5  $^{\circ}$ C.

The progress of the reaction can be monitored by TLC (EMD, 9. silica gel, grade 60,  $F_{254}$ ) with 1:1 EtOAc: hexanes and visualization with UV (starting material 1 has  $R_f = 0.45$  and the Boc-proline product 2 has  $R_f =$ 0.3); however, the end of reaction cannot be determined by TLC since the unreactive bis-tosylate carried forward from step A co-elutes with the monotosylate. NMR of the crude reaction mixture is uninformative due to broad peaks caused by Boc rotamers. Therefore, the end of reaction was assessed by deprotecting the Boc group and determining the amount of mono-tosylate that remained unreacted by <sup>1</sup>H NMR. A ~20 mg aliquot of the reaction mixture was evaporated then dissolved in 0.5 mL of CDCl<sub>3</sub> followed by addition of 0.2 mL of TFA. The sample was reacted for 15 min at room temperature then analyzed by <sup>1</sup>H NMR. The Ts-methyl group was diagnostic for assessing reaction completion: bis-tosylate,  $\delta$  2.42; product 3, δ 2.45; mono-Ts **1**, δ 2.52.

10. Given the broad peaks in the <sup>1</sup>H NMR spectrum (Note 12) the purity of the crude material from step B could not be estimated by NMR. The rough purity and yield estimates are based on the 65% recovery of material when subjected to flash chromatography in a separate experiment (Note 11).

11. Compound **2** (2.33 g crude weight) can be purified by column chromatography using 85 g silica gel (Fisher, 230-400 mesh, 60 Å) packed as a slurry with 2:1 hexanes: EtOAc, and eluted with 2:1 hexanes: EtOAc (600 mL), 1:1 hexanes:EtOAc (200 mL), and 1:2 hexanes: EtOAc (200 mL), taking 40 mL fractions. The desired product is obtained in fractions 16-23, ( $R_f = 0.3$ , 1:1 hexanes:EtOAc), which are combined and concentrated by rotary evaporation (40 °C, 20 mmHg) to give, after vacuum drying at room temperature to constant weight, 1.68 g of **2** (~90% purity, 65% recovery) as a pink foam.

12. At ambient temperature compound **2** is a mixture of 2 rotamers that cause broad peaks in the <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra. The following NMR data were collected at 360 K where the rotamers had partially coalesced. <sup>1</sup>H NMR (600 MHz, 360 K, DMSO-d<sub>6</sub>)  $\delta$ : 0.76 (br s, 1 H), 1.06 (br s, 1 H), 1.32 (s, 9 H, C(CH<sub>3</sub>)<sub>3</sub>], 1.39 (br s, 1 H), 1.67-1.73 (m, 1 H), 2.38 (s, 3 H, CH<sub>3</sub>), 2.74 (br s, 1 H), 3.08 (app q, *J* = 8.3 Hz, 1 H), 4.00 (dd, *J* = 8.9, 3.0, 1 H), 6.70 (d, *J* = 8.5 Hz, 1 H), 6.88 (d, *J* = 8.5 Hz, 1 H), 7.13-7.16 (m, 1 H), 7.20-7.25 (m, 3 H), 7.37 (d, *J* = 9.0 Hz, 1 H), 7.43-7.46 (m, 2 H), 7.48 (d, *J* = 8.3 Hz, 2 H), 7.88 (br d, *J* = 8.6 Hz, 1 H), 7.94 (d, *J* = 8.2 Hz, 1 H), 7.96-7.99 (m, 2 H), 8.08 (d, *J* = 8.8 Hz, 1 H), 8.52 (br s, 1 H, NH), 8.78 (br s, 1 H, NH). <sup>13</sup>C NMR (150 MHz, 360K, DMSO-d<sub>6</sub>)  $\delta$ : 20.4, 22.1, 27.6, 29.4, 45.8, 60.0, 78.5, 122.4, 123.7, 124.5, 124.7, 124.8, 125.3, 125.93, 125.95, 126.3, 127.4, 127.5, 128.3, 128.7, 128.9, 130.9, 131.0, 131.9, 132.1, 133.3, 134,5, 137.7, 142.5, 153 (br), 171.2.

13. The following reagents and solvents in Step C were used as received: trifluoroacetic acid (Sigma-Aldrich, >99%), dichloromethane (Fisher, ACS reagent, 99.5%), silica gel (Fisher, 230-400 mesh, 60 Å), EtOAc (Fisher, ACS reagent, 99%), and hexanes (Fisher, ACS reagent, >98.5%).

14. During the TFA addition, the temperature decreases from 22 °C to 19 °C.

15. Reaction progress can be monitored by TLC using 2:1 EtOAc:hexanes as eluent and visualized by UV. An aliquot of the reaction mixture is quenched into a mixture of 0.5 mL of 2N NaOH and 0.5 mL of dichloromethane with the bottom organic layer sampled for TLC.  $R_f$  product **3**, 0.3;  $R_f$  starting material, **2**, 0.8;  $R_f$  bis-tosylate, 0.9.

16. Addition of NaOH is exothermic and should be added at a rate to keep the internal temperature below 35 °C to prevent boiling of dichloromethane.

17. At the end of the NaOH addition, the pH is checked by pH paper and should be 8-10. If below 8, additional NaOH is added.

18. The organic layer is hazy due to the retention of a second phase water that is not completely removed upon drying with sodium sulfate.

19. A 6-cm diameter glass column is slurry-packed (2:1 EtOAc:hexanes) with silica gel (200 g). Crude product **3** co-mixed with silica is slurried in 2:1 EtOAc:hexanes and added to the top of the column. The column is topped with 0.5 cm of sand, then eluted with 2:1 EtOAc:hexanes (500 mL), 3:1 EtOAc:hexanes (500 mL), and EtOAc (1 L), taking 100 mL fractions. The chromatography is monitored by TLC (EtOAc,  $R_f$  0.5). The product elutes in fractions 9-15, which are combined and concentrated by rotary evaporation (40 °C water bath, 20 mmHg) in a 1-L flask, then transferred to a 250-mL round-bottomed flask for the final concentration. The white solid is dried under vacuum (20 mmHg) at room temperature for 28 h to constant weight (3.9 – 4.1 g).

20.  $(S_a,S)$ -*N*-[2'-(4-Methylphenylsulfonamido)-1,1'-binaphthyl-2-ylpyrrolidine-2-carboxamide (**3**) exhibits the following physical and spectroscopic properties:  $R_f$  0.5 (EtOAc);  $[\alpha]_D^{25}$  –95 (*c* 1.0, CHCl<sub>3</sub>); mp 196-197 °C, Lit<sup>7d</sup> 191-192 °C; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$ : 0.64–0.73 (m, 1 H), 1.15–1.28 (m, 2 H), 1.57–1.64 (m, 1 H), 1.74–1.84 (m, 1 H), 2.20–2.26 (m, 1 H), 2.35 (s, 3 H), 3.32 (dd, *J* = 4.0, 9.5 Hz, 1 H), 6.35 (br s, 1 H), 6.86 (d, *J* = 8.4 Hz, 1 H), 6.94 (d, *J* = 8.5 Hz, 1 H), 7.12 (d, *J* = 8.1, 2 H), 7.16– 7.22 (m, 2 H), 7.37–7.46 (m, 4 H), 7.87 (d, *J* = 7.9 Hz, 1 H), 7.95 (d, *J* = 8.2 Hz, 1 H), 8.00 (d, *J* = 9.1 Hz, 1 H), 8.06 (d, *J* = 9.0, 1 H), 8.19 (d, *J* = 9.0 Hz, 1 H), 8.82 (d, *J* = 9.0 Hz, 1 H), 9.31 (br s, 1 H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$ : 21.7, 25.4, 30.7, 46.3, 60.7, 117.0, 119.4, 119.6, 120.8, 124.3, 125.3, 125.4, 125.8, 127.6, 127.8, 128.3, 128.8, 129.7, 130.3, 130.7, 130.9, 131.4, 132.3, 132.7, 133.9, 135.9. 136.7, 144.1, 173.5; Anal. calcd. for C<sub>32</sub>H<sub>29</sub>N<sub>3</sub>O<sub>3</sub>S: C, 71.75; H, 5.46; N, 7.84; Found: C, 71.41; H, 5.15; N, 7.73.

21. The checkers determined the enantiomeric purity by SFC using a Lux-4 column (150 x 4.6mm, 5um particle size); isocratic elution, 40% MeOH with 25 mM *i*-butylamine/60% CO<sub>2</sub>; 3.0 mL/min flow; detection at 210 nm; 200 bar pressure; t<sub>r</sub> (*S*,*S*)= 4.5 min; t<sub>r</sub> (*R*,*R*)= 5.5 min; none of the enantiomer was detectable (ee >99%). The submitters determined enantiomeric purity by HPLC analysis at 254 nm using a Chiralpak AD-H column; isocratic elution, 80:20 hexanes: *i*-PrOH; 1mL/min: t<sub>r</sub> (*R*,*R*)= 51 min, t<sub>r</sub> (*S*,*S*)= 105 min. The (*R*,*R*)-enantiomer was prepared by the same procedure using (*R<sub>a</sub>*)-(-)-1,1'-binaphthyl-2,2'-diamine and Boc-*D*-proline

and exhibited the following physical properties: mp 195–197 °C;  $[\alpha]_D^{25}$  +93 (*c* 1.0, CHCl<sub>3</sub>).

22. The diastereomeric purity (de) was determined to be >99% by  ${}^{1}$ H NMR analysis in comparison to the  $(S_a, R)$ -diastereomer, which was prepared via the same procedure except that Boc-D-proline was used instead of Boc-L-proline. One aromatic proton in the  $(S_a, R)$ -diastereomer is upfield (6.54) ppm doublet) relative to the  $(S_a,S)$ -diastereomer (6.86 ppm doublet). The 6.54 ppm doublet was undetectible (<0.5%) in the  $(S_a,S)$ -diastereomer. The diastereomer  $(S_a, R)$ -N-[2'-(4-methylphenylsulfonamido)-1,1'-binaphthyl-2yl-pyrrolidine-2-carboxamide exhibits the following physical and spectroscopic properties: mp 134–137 °C, Lit<sup>7d</sup> 152-155 °C;  $[\alpha]_{D}^{25} + 6$  (*c* 1.0, CHCl<sub>3</sub>); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ: 1.50–1.55 (m, 2 H), 1.86–2.00 (m, 2 H), 2.30–2.34 (m, 1 H), 2.34 (s, 3 H), 2.57–2.62 (m, 1 H), 3.51 (dd, J = 4.5, 9.5 Hz, 1 H), 6.54 (d, J = 8.4 Hz, 1 H), 6.91 (d, J = 8.5 Hz, 1 H), 6.98 (dt, J= 1.1, 7.7 Hz, 1 H), 7.01 (d, J = 8.0 Hz, 2 H), 7.20–7.23 (m, 1 H), 7.34–7.41 (m, 4 H), 7.89 (dd, J = 8.3, 11.0 Hz, 2 H), 7.99 (d, J = 9.0 Hz, 1 H), 8.06 (d, J = 9.0 Hz, 1 H), 8.15 (d, J = 9.0 Hz, 1 H), 8.80 (d, J = 9.0 Hz, 1 H), 9.65 (br s, 1 H, NH); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ: 21.7, 26.1, 30.9, 47.0, 61.0, 117.6, 118.9, 120.0, 120.3, 124.7, 124.9, 125.6, 127.2, 127.3, 127.6, 128.3, 128.5, 129.8, 130.2, 130.7, 130.9, 131.2, 132.5, 132.9, 134.0, 135.8, 136.4, 144.0, 174.1.

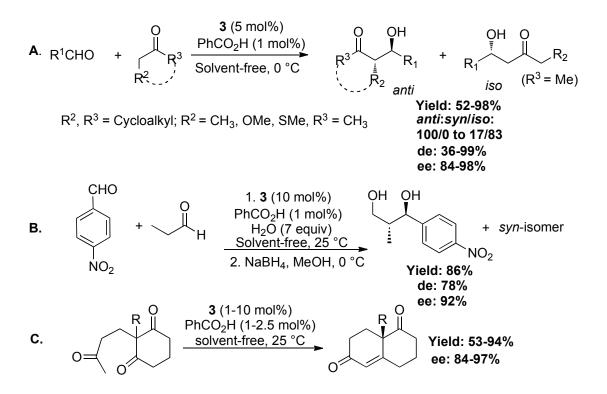
## Safety and Waste Disposal Information

All hazardous materials should be handled and disposed of in accordance with "Prudent Practices in the Laboratory"; National Academy Press; Washington, DC, 1995.

## 3. Discussion

The first generation of BINAM-prolinamides was introduced by several groups in 2006 to use in direct asymmetric aldol reactions<sup>5-7</sup> or other enantioselective processes.<sup>8</sup> ( $S_a$ ,S)-N-[2-(4-Methylphenylsulfonamido)-1,1'- binaphthyl-2'-yl]-pyrrolidine-2-carboxamide (**3**) is a novel BINAM-prolinamide-type organocatalyst<sup>5</sup> that was developed by Nájera's group<sup>9</sup> and others<sup>7d</sup> almost simultaneously. This ( $S_a$ )-binam-L-prolinamide sulfonamide derivative **3**<sup>9</sup> was designed by replacing one proline residue in the first generation catalyst<sup>7</sup> with an acidic sulfonamide group that could activate the

carbonyl group of the acceptor through hydrogen-bonding.<sup>10</sup> The efficiency of this catalyst when used with a small amount of benzoic acid as an additive has been proven in several aldol reactions, including the intermolecular aldol reaction between aldehydes and ketones (A, Scheme 1), the cross-aldol reaction between aldehydes (B, Scheme 1), and the intramolecular aldol reaction for the synthesis of the Wieland Miescher ketone (WMK) and related analogues (C, Scheme 1).



Scheme 1. Aldol processes catalyzed by N-tosyl-( $S_a$ )-binam-L-prolinamide 3

For all the processes, solvent-free reaction conditions could be applied using low catalyst loadings, obtaining the corresponding aldol products with good yields and diastereo- and enantioselectivities comparable to those achieved with other structurally similar catalysts<sup>11</sup> under different reaction conditions. For instance, the large-scale synthesis of the Wieland-Miescher ketone<sup>12</sup> requires only 1 mol% of catalyst **3** (see accompanying article).<sup>13</sup>

The preparation of catalyst **3** described here is a variant of those which already exist,  $^{9,10}$  and offers the following advantages:

1) Minimal purification steps: only the final product needs to be purified by chromatography.

2) The amide bond formation is efficiently accomplished using ethyl chloroformate, which avoids the use of SOCl<sub>2</sub> to form the acid chloride or the need for expensive coupling agents that, moreover, are difficult to remove at the end of the reaction.

This preparation can be also applied to obtain the enantiomer of the desired product,  $(R_a,R)$ -N- $[2'-(4-methylphenylsulfonamido)-1,1'-binaphthyl-2-yl]pyrrolidine-2-carboxamide, as well as the <math>(R_a,S)$ - and  $(S_a,R)$ -diastereomers.

- 1. Dpto. Química Orgánica and Instituto de Síntesis Orgánica, Universidad de Alicante, Apdo 99, E-03080 Alicante, Spain. Financial support from the MICINN (projects CTQ2007-62771/BQU, CTQ2010-20387 and Consolider INGENIO CSD2007-0006), FEDER, the Generalitat Valenciana (PROMETEO/2009/038), University of Alicante and European Community (COST Action CM0905: Organocatalysis (ORCA)).
- Laboratori de Química Orgànica, Facultat de Farmàcia, IBUB, Universitat de Barcelona, Av. Joan XXIII s/n, 08028-Barcelona, Spain. Financial support from the MICINN (projects CTQ2007-61338/BQU, CTQ2010-14846/BQU.
- **3.** The checker thanks Mirlinda Biba for measuring rotations, Zainab Pirzada for developing the chiral SFC assay, Bob Reamer for carrying out the high temperature NMR work on compound **2**, and WuXi Pharmatech for the preparative HPLC separation of the mono- and bistosylates.
- 4. Chen, T; Gao, J; Shi, M. Tetrahedron 2006, 62, 6289-6294.
- For general reviews of organocatalysis, see: (a) List, B. Chem. Commun.
   2006, 819-824. (b) Dondoni, A.; Massi, A. Angew. Chem. Int. Ed. 2008,
   47, 4638-4660. (c) Barbas III, C. F. Angew. Chem., Int. Ed. 2008, 47,
   42-47. (d) MacMillan, D. W. C. Nature 2008, 455, 304-308. (e)
   Bertelsen, S.; Jørgensen, K. A. Chem. Soc. Rev. 2009, 38, 2178-2189.
   (f) Grondal, C.; Jeanty, M.; Enders, D. Nature Chem. 2010, 2, 167-178.
- For reviews on the organocatalyzed direct aldol reaction, see: (a) Guillena, G.; Ramón, D. J. *Tetrahedron: Asymmetry* 2006, 17, 1465-1492. (b) Zlotin, S. G.; Kucherenko, A. S.; Beletskaya, I. P. *Russ. Chem.*

*Rev.* **2009**, *78*, 737-784. (c) Trost, B. M.; Brindle, C. S. *Chem. Soc. Rev.* **2010**, *39*, 1600-1632.

- For pioneering work using BINAM-prolinamides: (a) Gryko, D.; Kowalczyk, B.; Zawadzki, L. Synlett 2006, 1059-1062. (b) Guillena, G.; Hita, M. C.; Nájera, C. Tetrahedron: Asymmetry 2006, 17, 1493-1497.
   (c) Guizzetti, S.; Benaglia, M.; Pignataro, L.; Puglisi, A. Tetrahedron: Asymmetry 2006, 17, 2754-2760. (d) Ma, G.-N.; Zhang, Y.-P.; Shi, M. Synthesis 2007, 197-208.
- 8. (a) Xiong, Y.; Huang, X.; Gou, S.; Huang, J.; Wen, Y.; Feng, X. Adv. Synth. Catal. 2006, 348, 538-544. (b) Horillo Martínez, P.; Hultzch, K. C.; Hampel, F. Chem. Commun. 2006, 2221-2223.
- 9. Guillena, G.; Nájera, C.; Viózquez, S. F. Synlett 2008, 3031-3035.
- 10. For amide-based bifunctional organocatalysts in asymmetric reactions, see: Liu, X.; Lin, L.; Feng, X. *Chem. Commun.* 2009, 6145-6158.
- 11. (a) Guillena, G.; Hita, M. C.; Nájera, C.; Viózquez, S. F. *Tetrahedron: Asymmetry* 2007, *18*, 2300-2304. (b) Guillena, G.; Hita, M. C.; Nájera, C.; Viózquez, S. F. *J. Org. Chem.* 2008, *73*, 5933.
- 12. (a) Bradshaw, B.; Etxeberría-Jardí, G.; Bonjoch, J.; Guillena, G.; Nájera, C.; Viózquez, S. F. *Adv. Synth. Catal.* 2009, *351*, 2482-2490; (b) Bradshaw, B.; Etxeberría-Jardí, G.; Bonjoch, J. *J. .Am. Chem. Soc.* 2010, *132*, 5966–5967.
- 13. Bradshaw, B.; Etxeberría-Jardí, G.; Bonjoch, J.; Viózquez, S. F., Guillena, G.; Nájera, C. Org. Synth. 2011, 88, 330-341.

## Appendix Chemical Abstracts Nomenclature (Registry Number);

(*S<sub>a</sub>*)-(-)-1,1'-Binaphthyl-2,2'-diamine: (*S<sub>a</sub>*)-(-)-1,1'-Bi(2-naphthylamine); (18531-95-8)

*p*-Toluenesulfonyl chloride: Benzenesulfonyl chloride, 4-methyl-; (98-59-9) Pyridine; (110-86-1)

*N*-(*tert*-Butoxycarbonyl)-*L*-proline; (15761-39-4)

Ethyl chloroformate: Carbonochloridic acid, ethyl ester; (541-41-3)

Triethylamine; (121-44-8)

Trifluoroacetic acid; (76-05-1)



Josep Bonjoch was born in Barcelona (Catalonia, Spain) in 1952. He received his Ph.D. degree (1979) under the supervision of Prof. Joan Bosch at the University of Barcelona, Faculty of Chemistry. He then moved to the Faculty of Pharmacy at the same University, where he was promoted to Associate Professor (1984) and subsequently became Full Professor of Organic Chemistry in 1992. His main research involves the synthesis of complex nitrogen containing natural products, as a motive for developing new synthetic methodology.



Santiago Viózquez was born in Alicante (Spain) in 1981. He received his B.S. degree in chemistry at the Universidad de Alicante in 2006. He is now pursuing his Ph.D. at the Universidad de Alicante under the supervision of G. Guillena and C. Nájera. His research concerns asymmetric organocatalysis with prolinamides derivatives.



Gabriela Guillena received her BSc degree (1993) from University of Alicante. After spending one year as postgraduate student in the group of D. Seebach at the ETH (Zurich), she returned to University of Alicante and received her MSc (1995) and PhD (2000) degrees under the supervision of C. Nájera. After two years as a postdoctoral fellow at research group of G. van Koten (University of Utrecht, Netherlands), she returned to the University of Alicante where she became Assistant Professor in 2003 and Associate Professor in 2008. Her current research interests are focused on new organic methodologies and asymmetric organocatalysis.



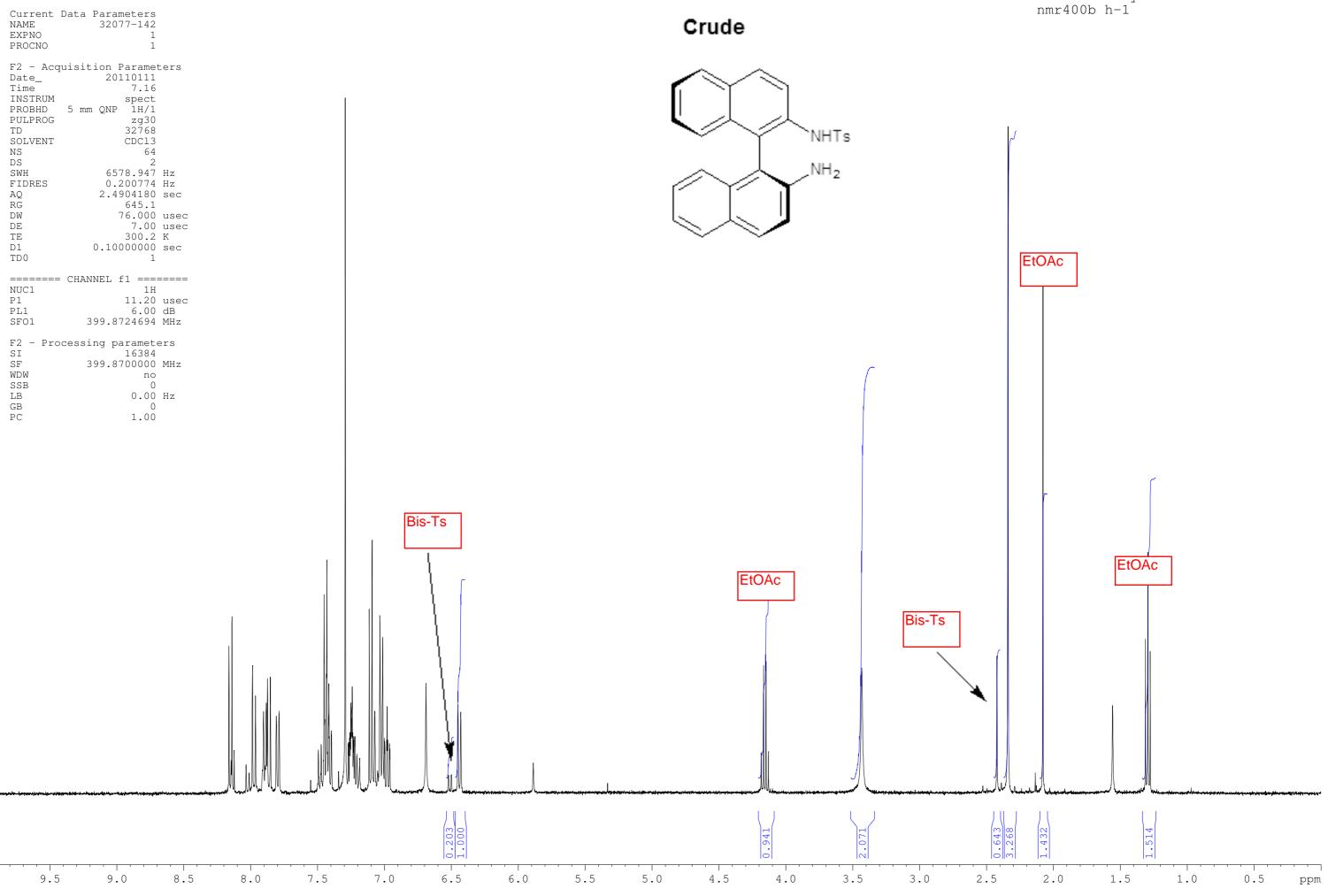
Carmen Nájera obtained her B.Sc. (1973) from University of Saragossa and her PhD (1979) at the University of Oviedo under the supervision of J. Barluenga and M. Yus. She performed her postdoctoral work at the ETH (Zurich) with D. Seebach, at the Dyson Perrins Laboratory (Oxford) with J. E. Baldwin, at Harvard University with E. J. Corey, and at Uppsala University with J.-E. Bäckvall. She became Associate Professor in 1985 at the University of Oviedo and Full Professor in 1993 at the University of Alicante. Her scientific contributions are focused on synthetic organic chemistry such as sulfone chemistry, new peptide coupling reagents, oximederived palladacycles, asymmetric metal catalysis and organocatalysis.



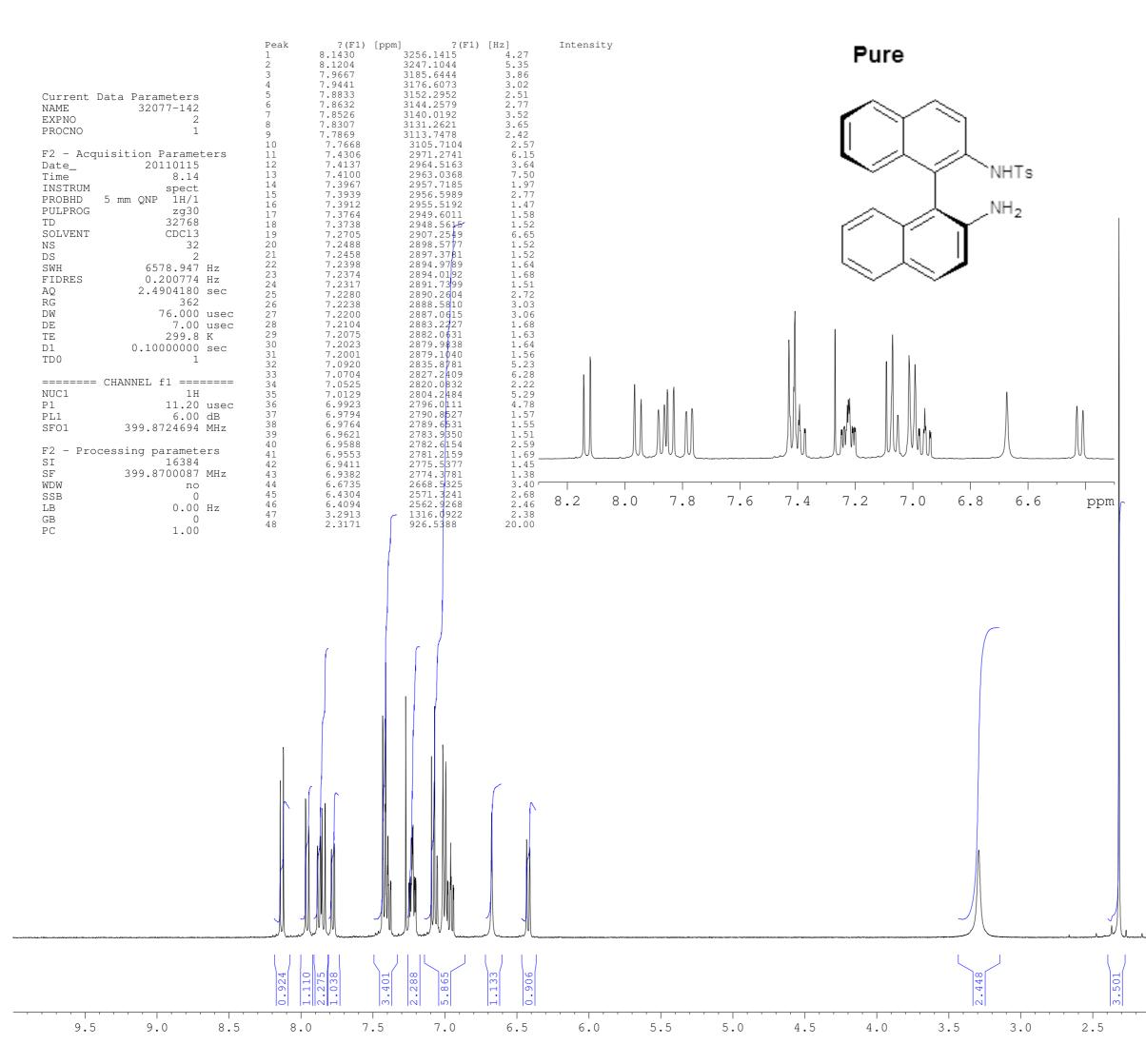
Ben Bradshaw was born in 1974 in Southport, England. He studied Chemistry at the University of Manchester, where he obtained his PhD in 2001 under the supervision of Professor John Joule. After postdoctoral work with Professor Jim Thomas on the total synthesis of the Bryostatins he joined the group of Professor Josep Bonjoch at the University of Barcelona. In 2008 he was promoted to the position of assistant professor where his research interests include the application of organocatalysis to the total synthesis of complex natural products.



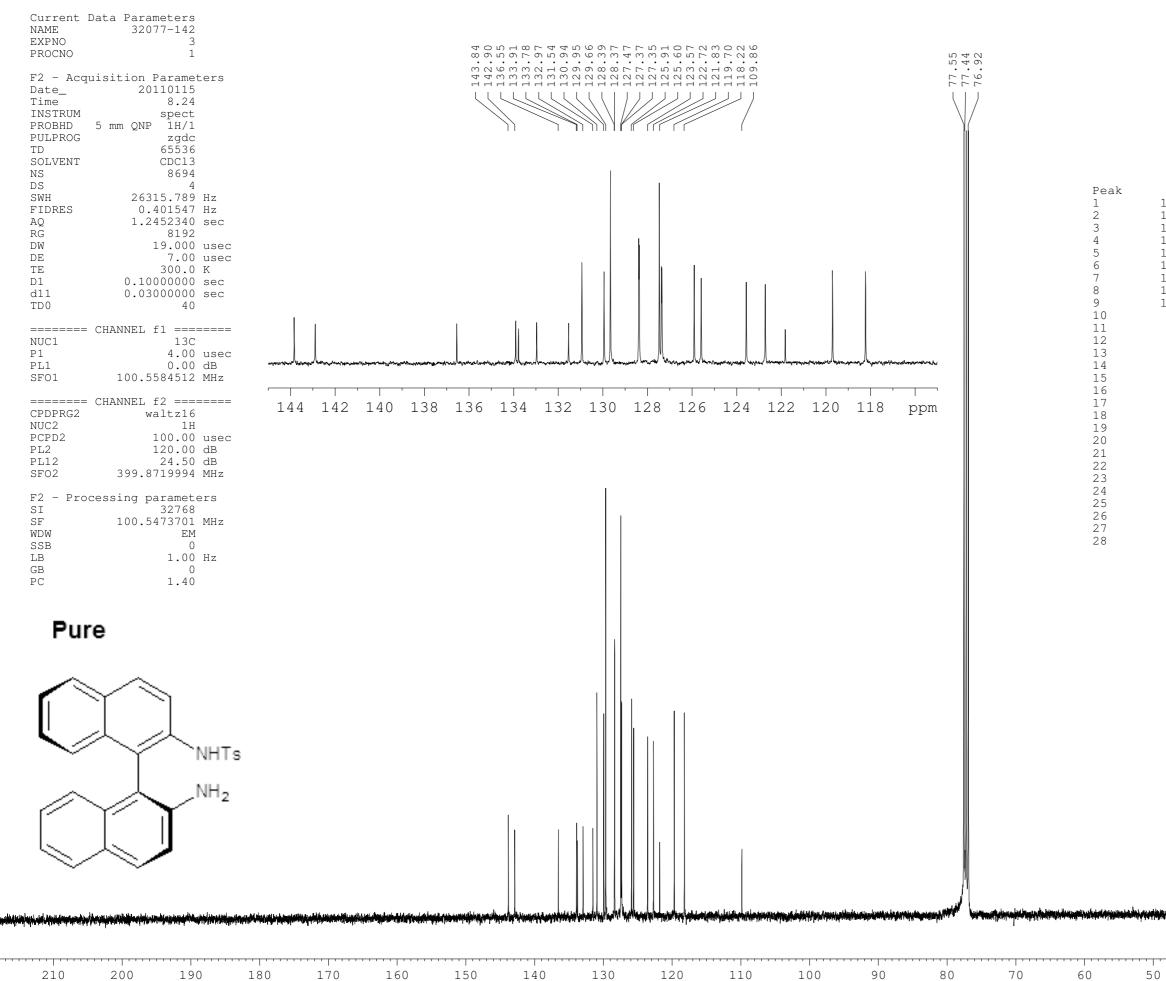
Gorka Etxebarria-Jardí was born in 1981 in Barcelona, Catalonia. He obtained his BSc in Chemistry (2004) and MSc in synthesis of antiretroviral nucleoside drugs (2005) from the University of Barcelona. In 2006, he joined the research group of Prof. Josep Bonjoch and is currently completing his Ph.D in asymmetric catalysis and natural product synthesis.



32077-142 S-Binam-tosylate nmr400b h-1



mono-tosylate
WuXi separation
nmr400b h-1



	21.7		
?(F1) [ppm]	?(F1)	[Hz]	Intensity
143.8413	14462.8644	1.32	
142.8954	14367.7567	1.22	
136.5498	13729.7233	1.14	
133.9067	13463.9665	1.23	
133.7841	13451.6394	1.02	
132.9706	13369.8441	1.18	
131.5401	13226.0111	1.17	
130.9423	13165.9039	2.83	
129.9503	13066.1609	2.62	
129.6632	13037.2938	5.43	
128.3936	12909.6388	3.50	
128.3718	12907.4469	3.33	
127.4698	12816.7532	5.13	
127.3742	12807.1408	2.76	
127.3513	12804.8383	2.81	
125.9097	12659.8892	2.77	
125.5958	12628.3274	2.52	
123.5735	12424.9904	2.41	
122.7197	12339.1431	2.39	
121.8282	12249.5051	0.99	
119.7032	12035.8420	2.63	
118.2250	1887.2128	2.64	
109.8580	1045.9330	0.91	
77.5539	7797.8407	14.58	
77.4378	7786.1671	0.89	
77.2359	7765.8666	15.00	
76.9179	7733.8926	14.69	
21.7153	2183.4163	2.02	

mono-tosylate WuXi separation nmr400b c-13

72

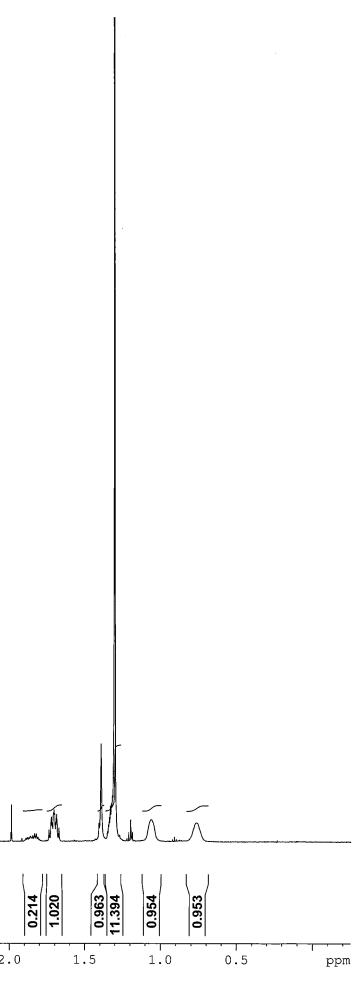
40

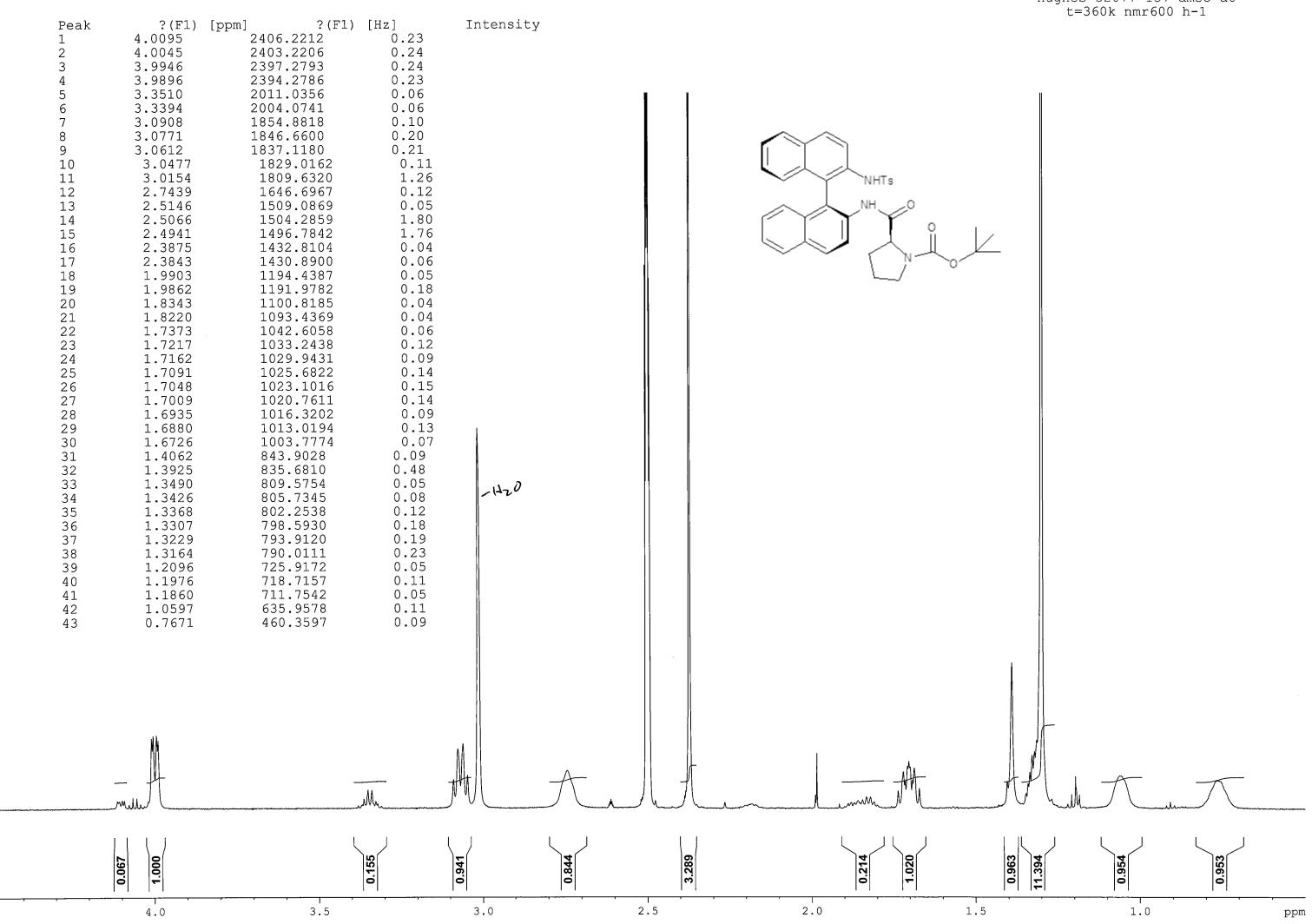
30

10

NAME EXPNO PROCNO F2 - Ac Date_ Time INSTRUM PROBHD PULPROG TD SOLVENT NS DS SWH FIDRES AQ RG DW DE TE D1 TDO ======= NUC1 P1 PL1 SFO1	5 mm Multinucl zg30 65536	c ec ec c == ec z				HTs H								
SSB LB GB PC	0.00 Hz 0 1.00									M	t			
	0.988	1.057 3.278 0.921	4.379 1.033 3.270 1.112	1.090					0.067	0.155	0.941	0.844	3.289	
9.5	9.0 8.5	8.0	7.5	7.0 6.	5 6.0	5.5	5.0	4.5	4.0	3.5	3.0		2.5	2.(

## hughes 32077-137 dmso-d6 t=360k nmr600 h-1

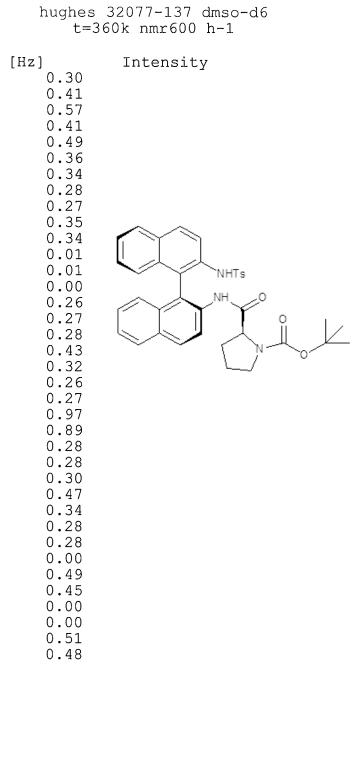




hughes 32077-137 dmso-d6

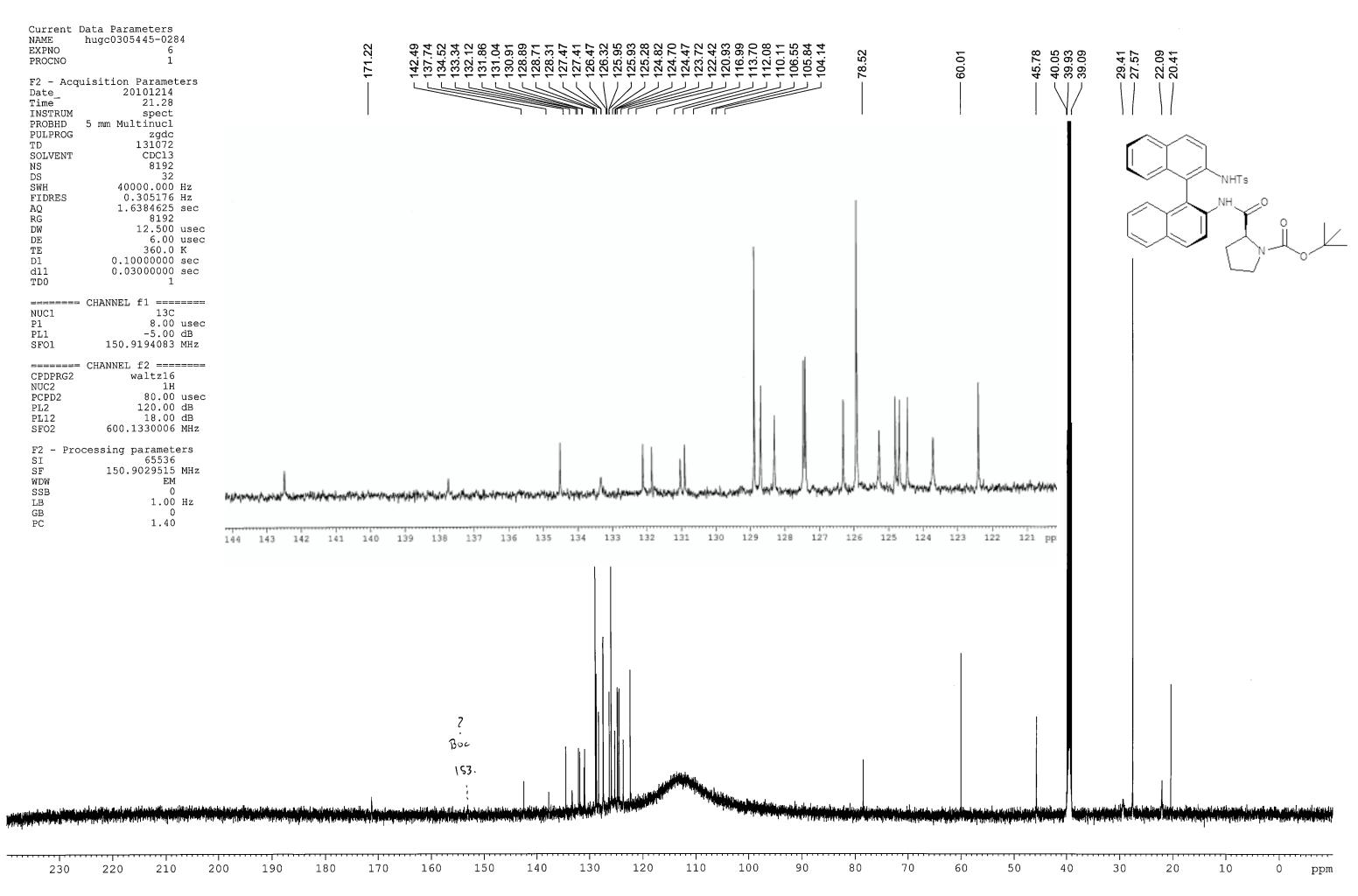
Peak	?(F1)	[ppm] ?(F1)	[Hz]	Intensity			
1	12.1509	7292.1196	0.00		<b>D</b> 1	0 (71)	
Current Data Parameters $\overline{2}$	12.0740	7245.9696	0.00		Peak	?(F1) [pp	
NAME hugc0305445-0284	12.0514	7232.4067	0.01		37	7.4526	4472.5288
EXPNO 5	11.9846	7192.3180	0.00		38	7.4494	4470.6084
PROCNO 1 4 5	8.9008	5341.6371	0.00		39	7.4475	4469.4682
5 F2 - Acquisition Parameters					40	7.4454	4468.2079
Date 20101214	8.8310	5299.7480	0.00		41	7.4409	4465.5073
Time 17.19 /	8.7763	5266.9209	0.25		42	7.4358	4462.4467
INSTRUM spect 8	8.6937	5217.3502	0.00		43	7.4341	4461.4264
PROBHD 5 mm Multinucl 9 PULPROG zg30 10	8.6899	5215.0697	0.00		44	7.4291	4458.4258
$\begin{array}{ccc} & & & & & & \\ \text{TD} & & & & & 65536 \end{array} 10$	8.5244	5115.7482	0.39		45	7.4277	4457.5856
SOLVENT DMSO 11	8.2212	4933.7888	0.01		46	7.3775	4427.4591
NS 32 12	8.1192	4872.5755	0.01				
DS 4 12376.237 Hz	8.0871	4853.3113	0.53		47	7.3626	4418.5171
FIDRES 0.188846 <b>1</b> 24	8.0724	4844.4894	0.61		48	7.3388	4404.2340
AQ 2.6477449 sec	8.0269	4817.1835	0.01		49	7.3350	4401.9536
NG 52 1 G	7.9880	4793.8384	0.53		50	7.2935	4377.0482
DW 40.400 tsec DE 6.00 lsec	7.9797		0.60		51	7.2524	4352.3828
DE 6.00 11.5/ec TE 360.0 14.8		4788.8574			52	7.2507	4351.3626
D1 0.10000000 Sec TD0 1	7.9746	4785.7967	0.59		53	7.2410	4345.5413
TD0 1 19	7.9649	4779.9754	0.59		54	7.2387	4344.1610
20	7.9506	4771.3936	0.49		55	7.2365	4342.8407
======= CHANNEL f1 ====27== NUC1 1H 22	7.9370	4763.2318	0.51		56	7.2268	4337.0195
P1 9.00 <del>65</del> ec	7.8931	4736.8861	0.21		57	7.2251	4335.9993
PL1 0.00 2B	7.8788	4728.3042	0.18		58	7.2142	4329.4578
SFO1 600.1337058 Мндг	7.8547	4713.8411	0.01		59	7.2008	4321.4161
F2 - Processing parameters	7.8418	4706.0994	0.01		60	7.1623	4298.3111
SI 32768 26	7.8284	4698.0577	0.01				
SF 600.1300003 MHz	7.8032	4682.9344	0.00		61	7.1604	4297.1709
WDW no oo	7.6237	4575.2111	0.00		62	7.1510	4291.5296
SSB 0.28 LB 0.00 😥	7.6125	4568.4896	0.00		63	7.1486	4290.0893
GB 0 30	7.5803	4549.1654	0.00		64	7.1463	4288.7090
PC 1.00 31	7.5629	4538.7232	0.00		65	7.1368	4283.0078
					66	7.1350	4281.9276
32	7.5307	4519.3990	0.00		67	7.0783	4247.9002
33	7.4898	4494.8537	1.15		68	6.8884	4133.9355
34	7.4760	4486.5719	1.27		69	6.8742	4125.4136
35	7.4607	4477.3899	0.30		70	6.8429	4106.6296
36	7.4591	4476.4297	0.32		71	6.7378	4043.5559
					72	6.7118	4027.9525
					73	6.6977	4027.9525
					15	0.0977	4019.4907

0.988 3.278 3.270 1.112 1.057 1.057 0.921 4.379 1.033 1.082 8.5 7.5 8.0 7.0 9.0

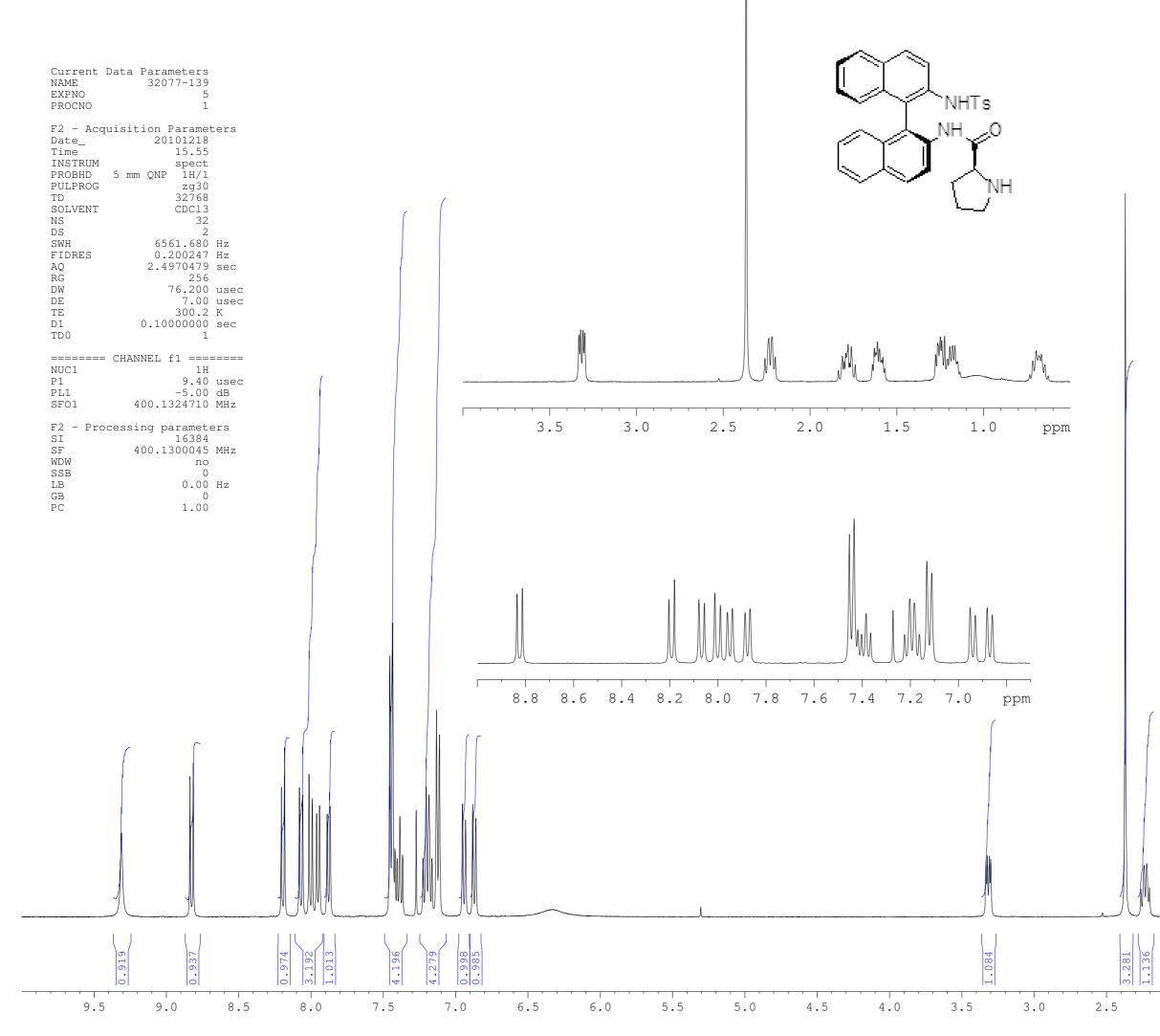


6.5

1.090



hughes 32077-137 dmso-d6 t=360k nmr600 c-13

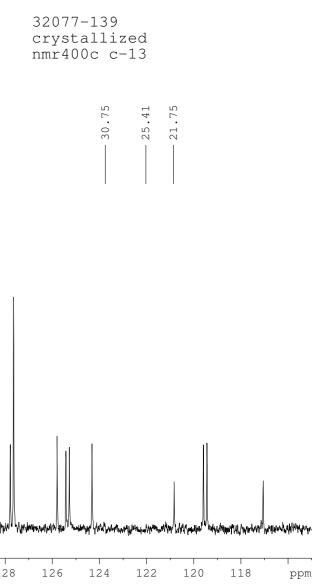


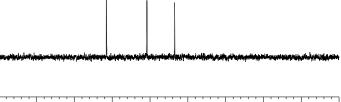
#### 32077-139 crystallized nmr400c h-1

Pea 1 2 3 4 5 6 7 8 9 10 11 2 2 3 4 5 6 7 8 9 10 11 2 2 2 3 4 5 6 7 8 9 10 11 2 2 2 3 2 4 5 6 7 8 9 10 11 2 2 2 3 2 4 5 6 7 8 9 10 11 2 2 2 3 2 4 5 6 7 8 9 10 11 2 2 2 3 2 4 5 6 7 8 9 10 11 12 2 2 3 2 4 5 6 7 8 9 10 11 12 2 2 3 2 4 5 6 7 8 9 10 11 12 2 2 3 2 4 5 6 7 8 9 0 11 12 2 2 3 2 4 5 6 7 8 9 0 11 12 2 2 3 2 4 5 6 7 8 9 0 11 12 2 2 3 2 4 5 6 7 8 9 0 11 12 2 2 3 2 4 5 6 7 8 9 0 11 12 2 2 3 2 4 5 6 7 8 9 0 11 12 2 2 3 2 4 5 6 7 8 9 0 11 1 2 2 2 3 2 4 5 6 7 8 9 0 11 1 2 2 2 3 2 4 5 6 7 8 9 0 11 1 2 2 2 3 2 4 5 6 7 8 9 0 12 2 2 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3	9 8 8 8 8 8 8 8 7	?(F1) .3092 .8353 .8127 .2035 .1809 .0788 .0562 .0124 .9897 7.9305 7.8868 7.8664 7.4534 7.4534 7.4534 7.4181 7.4028 7.837 7.3652 7.2711 7.2018 7.1827 7.3652 7.2711 7.104 6.9506 6.9294 6.8589 3.3229 3.32991 3.22991 3.22991 3.3229 3.32991 3.22991 3.3229 3.32991 3.22991 3.3291 3.3291		2 3 3 3 2 3 3 2 3 2 3 2 3 2 3 2 3 2 3 2 3 3 2 3 3 2 3 3 2 3 3 2 3 3 3 2 3 3 3 2 3 3 3 3 3 3 3 3 3 3 3 3 3	2 3 7 7 3 4 4 9 5 3 1 3 8 2 2 0 4 6 9 9 4 7 3 0 2 2	2] 2.31 3.91 4.20 3.68 4.68 3.58 4.01 3.26 2.86 3.07 7.226 8.16 1.90 1.64 1.71 2.94 3.61 3.09 1.62 5.08 3.13 2.71 1.72 1.54 1.70 1.61 0.77 1.42 1.61 0.77 1.42 1.61 0.77 1.42 1.61 0.77 1.42 1.61 0.77 1.42 1.61 0.57 1.09 0.57 1.09 0.65 0.57 1.09 1.08 0.057 1.09 1.08 0.57 1.09 1.08 0.57 1.09 1.08 0.57 1.09 1.08 0.57 1.09 1.08 0.57 1.09 1.08 0.57 1.09 1.08 0.57 1.09 1.08 0.57 1.09 1.08 0.57 1.09 1.08 0.57 1.09 1.08 0.57 1.09 1.08 0.57 1.09 1.08 0.57 1.09 1.08 0.57 1.09 1.08 0.57 1.09 1.08 0.57 1.09 1.08 0.57 1.09 1.08 0.057 1.09 1.08 0.057 1.09 1.08 0.057 1.09 1.08 0.057 1.09 1.08 1.	In
	1.138	1.180	2.146	(	1.000		
2.0	. , 1	1.5		1.0	0.5	)	ppm

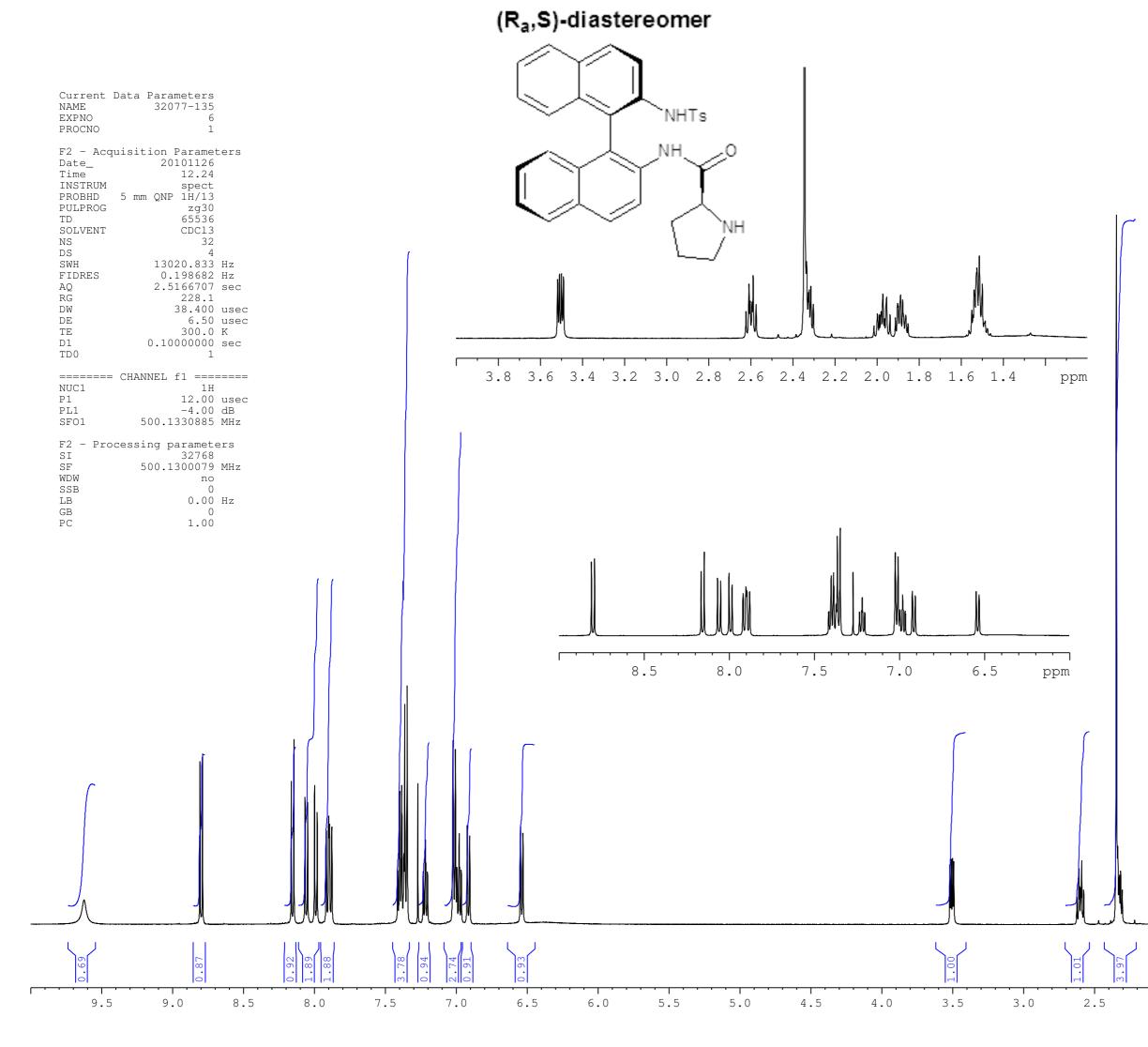
Inter

Current Data Parameters NAME 32077-139 EXPNO 6 PROCNO 01 F2 - Acquisition Parameters Date_ 20101218 Time 16.03 INSTRUM spect PROBHD 5 mm QNP 1H/1 PULPROG zgdc TD 65536 SOLVENT CDC13	4 0 U 4 0 0 1 0 0	132.32 132.32 132.32 132.32 130.30 130.30 129.69 128.81 128.81 128.33 129.69 125.42 125.42 125.42 125.42 125.43 117.04	77.56	60.67
NS       2006         DS       4         SWH       28248.588 Hz         FIDRES       0.431039 Hz         AQ       1.1600549 sec         RG       8192         DW       17.700 usec         DE       7.00 usec         TE       300.2 K         D1       0.1000000 sec         d11       0.3000000 sec         TD0       40         ==================================	Peak?(F1)[ppm]1173.53302144.08363136.71744135.87425133.91836132.71977132.31878131.39419130.697611130.295912129.687713128.811114128.333215127.778816127.637917125.789418125.417419125.271920124.308821120.822622119.578723119.429824117.04292577.55562677.23772776.91992860.67002946.28533030.74933125.41263221.7450	1       ?(F1) [Hz]       Intensity         17459.6321       0.91         14496.6470       0.79         13755.5134       0.77         13670.6767       0.89         13473.8882       0.89         13353.2938       0.79         13312.9481       0.73         13219.9215       0.80         13171.6677       0.78         13149.8448       1.40         13109.4286       1.42         13048.2359       2.98         12960.0388       1.26         12911.9560       1.41         12856.1763       1.31         12841.9999       3.50         12656.0173       1.40         12618.5893       1.24         12603.9502       1.29         12507.0500       1.32         12156.2939       0.77         12031.1417       1.30         12016.1604       1.32         11776.0079       0.77         7803.0821       9.86         7771.0973       10.00         7739.1226       9.74         6104.1755       1.65         4656.8912       1.45         3093.7716       1.57		
Manandara and a na an a				





			1		1				
5	40	35	30	25	20	15	10	5	ppm



### 32077-135

recrystallized R-Binam-Ts-L-pro nmr500c h-1

Peak 1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 21 22 33 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 21 22 33 4 25 26 27 28 9 30 31 22 33 4 5 6 7 8 9 00 11 12 13 14 15 16 7 8 9 20 21 22 33 4 25 26 27 28 9 30 31 22 33 4 5 6 6 7 8 9 00 11 22 23 24 25 26 27 28 9 30 31 22 33 4 5 5 6 7 8 9 00 11 22 23 24 25 26 27 28 9 30 31 22 33 4 5 5 6 6 7 8 9 9 00 11 22 23 24 25 26 27 28 9 30 31 22 33 4 5 5 6 6 7 8 9 9 0 1 22 23 24 25 26 27 28 9 30 31 2 33 4 5 5 6 6 7 8 9 9 0 1 22 23 24 25 26 27 28 9 30 31 2 33 4 5 5 6 6 7 8 9 9 0 1 22 3 34 5 5 6 6 7 8 9 9 0 1 22 3 34 5 5 6 7 8 9 9 0 1 2 2 3 7 8 9 9 0 1 2 2 3 2 4 2 5 2 6 7 28 9 9 0 1 2 2 3 2 4 5 5 6 7 7 8 9 0 0 1 2 2 3 34 5 5 6 7 7 8 9 0 0 1 2 5 3 4 5 5 6 7 7 8 9 0 0 1 2 5 5 6 7 7 8 9 0 0 1 2 5 5 5 6 7 7 8 9 0 0 1 2 5 5 5 5 5 7 5 8 9 0 0 1 2 5 5 5 5 5 7 5 8 9 0 0 1 2 5 5 5 5 5 7 5 8 9 0 0 1 2 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	2 (F1) 8.8073 8.7892 8.1626 8.0486 7.9993 7.9812 7.9166 7.9001 7.8946 7.8780 7.4139 7.3429 7.3463 7.3629 7.3463 7.3702 7.2335 7.2313 7.2171 7.2145 7.2008 7.0224 7.0063 6.9955 6.9922 6.9768 6.9935 6.9922 6.9768 6.9955 6.9922 6.9768 6.9650 3.5188 3.5000 3.4910 2.6232 2.6090 2.6035 3.5188 3.5000 3.4910 2.6232 2.6090 2.6035 3.5188 3.5000 3.4910 2.6232 2.6090 2.6035 3.5188 3.5000 3.4910 2.6232 2.6090 2.6035 3.5188 3.5000 3.4910 2.6232 2.6090 2.6035 3.5188 3.5000 3.4910 2.6232 2.6090 2.6035 3.5188 3.5000 3.4910 2.6235 3.5188 3.5000 3.4910 2.6235 3.5188 3.5000 3.4910 2.6235 3.5188 3.5000 3.4910 2.6235 3.5188 3.5000 3.4910 2.6235 3.5188 3.5000 3.4910 2.6235 3.5188 3.5000 3.4910 2.6235 3.5188 3.5000 3.4910 2.6235 3.5188 3.5000 3.4910 2.6235 3.5188 3.5000 3.4910 2.6235 3.5188 3.5000 3.4910 2.6235 3.5188 3.5000 3.4910 2.6235 3.5188 3.5000 3.4910 2.6235 3.5188 3.5000 3.4910 2.6235 3.5188 3.5000 3.4910 2.6235 3.5188 3.5000 3.4910 2.6235 3.5188 3.5000 3.4910 2.6235 3.5188 3.5000 3.4910 2.6235 3.5188 3.5000 3.4910 3.5071 1.9246 1.9385 1.93571 1.9022 1.8991 1.88731 1.9022 1.8911 1.4995 3.5141 1.4995 3.5141 3.4956 3.5141 3.4956 3.5141 3.4956 3.5141 3.4956 3.5141 3.4956 3.5141 3.4956 3.5141 3.4956 3.5141 3.5279 3.52	<pre>[ppm] ?(F1) 4404.7950 4395.7427 4082.3612 4073.3088 4034.3987 4025.3464 4000.6900 3991.6376 3959.3292 3951.0771 3948.3264 3940.0242 3707.9139 3707.1137 3700.3619 3693.4601 3663.66583 3685.6081 3682.4072 3674.1051 3636.0452 3617.6904 3616.5901 3610.7886 3609.4883 3608.1879 3602.4364 3601.3362 3512.1130 3504.0609 3498.7095 3497.6592 3490.5074 3489.3070 3483.4055 3462.3052 3462.1000 3453.5978 3274.5012 3266.0990 1759.8575 1755.3563 1750.4550 1745.9539 1311.9410 304.8392 1302.0885 1297.7874 1294.9366 1287.8848 1172.7549 1167.9536 1267.8848 1172.7549 1167.9536 1267.8848 1172.7549 1167.9536 1267.8848 1172.7549 1167.9536 1267.8848 1172.7549 1167.9536 1267.8848 1172.7549 1167.9536 1267.8848 1172.7549 1167.9536 1267.8848 172.7549 1167.9536 1287.8848 172.7549 1167.9536 1287.8848 172.7549 1167.9536 1287.8848 172.7549 1167.9536 1287.8848 172.7549 1163.7025 1161.8020 1157.8010 1151.8494 99.5598 994.7086 990.4075 986.6065 977.5541 969.5020 951.3473 949.7969 944.4955 939.3423 775.0015 771.9507 769.4500 764.1486 762.6482 757.2468 749.9449 </pre>	<pre>[Hz] II 4.58 4.85 4.19 5.28 3.59 3.42 3.90 3.16 2.63 3.04 2.82 2.76 1.48 1.52 3.75 3.92 1.91 1.99 6.18 6.73 4.10 1.49 1.49 1.57 2.43 1.74 1.49 1.56 2.55 1.67 1.54 1.44 2.81 2.49 2.80 2.56 1.70 1.57 1.07 1.64 1.56 2.55 1.67 1.57 1.07 1.07 1.04 1.88 1.86 1.78 0.79 1.57 1.07 1.07 1.04 1.51 0.99 0.71 0.69 1.00 0.93 1.24 1.10 0.67 0.77 1.27 1.19 0.69 1.00 0.93 1.24 1.10 0.67 0.82 0.68 1.40 1.28 1.88 1.86 1.75 1.07 1.07 1.07 1.07 1.07 1.04 1.33 1.40 1.51 0.99 0.71 0.69 1.00 0.93 1.24 1.10 0.67 0.82 0.68 1.40 1.28 1.88 1.56</pre>
2.0	<u>1.5</u>	······································	 5 ppm

Int