

A Publication of Reliable Methods for the Preparation of Organic Compounds

Working with Hazardous Chemicals

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September 2014: The paragraphs above replace the section "Handling and Disposal of Hazardous Chemicals" in the originally published version of this article. The statements above do not supersede any specific hazard caution notes and safety instructions included in the procedure.





Submitted by Gene W. Wong, Tyler T. Adint, Clark R. Landis.¹ Checked by Neil Strotman, James Cuff and David Hughes.

1. Procedure

Caution! Carbon monoxide is a highly toxic gas and manipulations should be conducted in a well-ventilated fume hood in the vicinity of a carbon monoxide detector. Hydrogen gas is highly flammable and explosive gas. Precautions should be taken when using synthesis gas (H_2 /CO mixtures).

A. Allyl (t-butyldimethyl)silyl ether 1. An oven-dried three-necked, 500mL round-bottomed flask equipped with a 3-cm PTFE-coated oval stir bar is charged with allyl alcohol (8.6 g, 0.15 mol, 1.0 equiv), imidazole (21.1 g, 0.31 mol, 2.1 equiv), and dimethylformamide (100 mL). (Note 1) The flask is fitted with a gas inlet adapter connected to a nitrogen line and a gas bubbler. The other two necks are capped with rubber septa; a thermocouple probe is inserted through one of the septa. (Note 2) The stirred solution is cooled to 4 °C with an ice-water bath, then *t*-butyldimethylsilyl chloride (25.5 g, 0.17 mol, 1.1 equiv) is added in portions over 2 min. (Note 3) The water bath is removed and the hazy solution is stirred 15 h at 22–23 °C. (Note 4) The reaction mixture is transferred to a 500-mL separatory funnel along with hexanes (200 mL) and water (60 mL). After mixing and settling, the aq. layer is removed and the organic layer washed with brine (75 mL). The hazy organic layer is filtered through a bed of sodium sulfate (50 g) in a medium porosity sintered glass funnel into a 500-mL round-bottomed flask, using hexanes (2 x 50 mL) to rinse the filter cake. The filtrate is concentrated by rotary evaporation (40 °C, 20 mmHg) to an oil (22 g). Since a substantial portion of the product co-distills during concentration, the distillate from the receiver flask is re-concentrated, providing additional product (2.8 g). The combined crude product is purified by column chromatography (Note 5) to afford allyl (*t*-butyldimethyl)silyl ether **1** (18.1–18.5 g, 71–73% yield) as a colorless oil. (Note 6)

B. (2*R*)-3-[[(1,1-dimethylethyl)dimethylsilyl]oxy]-2-methylpropanal 2. In a glovebox, bis[(S,S,S)-DiazaPhos-SPE] (9.2 mg, 0.0070 mmol, 0.024 mol%) is added to a 20 mL vial followed by 0.22 mL CDCl₃. To a separate 4 mL vial is added Rh(acac)(CO)₂ (10 mg) followed by 1.94 mL toluene to prepare a 20 mM stock solution; 0.29 mL (1.5 mg, 0.0058 mmol, 0.020 mol%) of this stock solution is transferred to the 20 mL vial containing the ligand solution (Notes 7, 8). Allyl (t-butyldimethyl)silyl ether 1 (5.00 g, 6.17 mL, 29 mmol, 1.0 equiv) is added to the vial. The solution is divided into 4 x 4 mL vials, each containing a 5-mm glass bead, and loaded into a Symyx Heated Orbital Shaker System (HOSS) contained within the glove box. (Note 9) The system is taken through 3 cycles of pressurization (150 psig of 1:1 H₂:CO)/depressurization (0 psig) to replace the nitrogen atmosphere with synthesis gas. The system is pressurized with 150 psig 1:1 H₂:CO and heated to 60 °C for 16 h. (Note 10) The mixture is cooled and the system depressurized, then the vials are removed from the glove box. (Note 15) The hydroformylation reaction mixture is purified immediately (Note 16) by flash column chromatography (Note 17) to afford colorless oils of branched isomer (R)-2 (3.24-3.43 g, 55-58% yield, Notes 18-20) and linear isomer **3** (1.62–1.70 g, 27–29% yield, Note 21).

2. Notes

1. The following reagents and solvents in step A were used as received: allyl alcohol (Sigma-Aldrich), imidazole (Acros, 99%), *t*-butyldimethylsilyl chloride (Acros, 98%), anhydrous DMF (Sigma-Aldrich, 99.8%), hexanes (Fisher, ACS reagent, >98.5%), ethyl acetate (Fisher, ACS reagent, >99.5%) and silica gel (Fisher, 230-400 mesh, 60 Å).

2. The internal temperature is monitored using a J-Kem Gemini digital thermometer with a Teflon-coated T-Type thermocouple probe (12-inch length, 1/8 inch outer diameter, temperature range -200 to +250 °C).

3. The reaction warmed to 25 °C over 5 min after addition of TBS-Cl.

4. The reaction was monitored by TLC, 5% EtOAc/hexanes, $R_{\rm f}$ 0.6, $KMnO_4\,stain.$

5. Silica gel (250 g) was slurry-packed in a 5-cm diameter column using 2.5% EtOAc/hexanes. The product was eluted with 2.5% EtOAc/hexanes, collecting 100 mL fractions. Fractions 5-13 were combined and concentrated by rotary evaporation (40 °C, 20 mmHg) to afford 1 (14.5–15.4 g) as a colorless oil. The distillate from concentration was re-concentrated to provide an additional 2.7–4.0 g (combined yield, 18.1–18.5 g, 71–73%). The distillate from the final product concentration was assayed by ¹H NMR using toluene as an internal standard, indicating 2.4 g (10% yield) 1 was present in the distillate.

6. *Allyl (t-butyldimethyl)silyl ether 1* has the following physical and spectroscopic data: ¹H NMR (400 MHz, CDCl₃) δ : 0.09 (s, 6 H), 0.93 (s, 9 H), 4.18–4.20 (m, 2 H), 5.07–5.11 (m, 1 H), 5.25–5.30 (m, 1 H), 5.89–5.97 (m, 1 H); ¹³C NMR (100 MHz, CDCl₃) δ : –5.0, 18.6, 26.2, 64.3, 114.1, 137.8; GC-MS (EI) *m/z*: 172 (6 %) [M⁺], 157 (6%), [M - CH₃], 116 (31%), 115 (100%) [M - *t*-Bu], 99 (21%), 85 (69%) [M – Me₂,*t*-Bu], 75 (28%), 59 (48%); GC purity: 98% (t_R = 4.4 min; conditions: Agilent DB35MS column; 30 m x 0.25 mm; initial temp 60 °C, ramp at 20 °C/min to 280 °C, hold 15 min).

7. The following reagents and solvents in step B were used as received by the checkers: dicarbonylacetylacetonato rhodium(I) (Strem), toluene (Sigma Aldrich, anhydrous, >99.9%), CDCl₃ (Sigma-Aldrich, 99.8% atom % D), SynGas (49% carbon monoxide/51% hydrogen, Airgas) and bis[(S,S,S)-DiazaPhos-SPE] ligand: 2,2',2",2'''-(1,2-phenylenebis[(1S,3S)tetrahydro-5,8-dioxo-1*H*-[1,2,4]diazaphospholo[1,2-a]pyridazine-2,1,3(3*H*)triyl])tetrakis(N-[(1S)-1-phenylethyl])benzamide (Sigma-Aldrich). The ligand was prepared by the submitters according to their published procedure.² The submitters recrystallized dicarbonylacetylacetonato rhodium(I) from toluene and hexanes as fine green crystals. CHCl₃ may be used instead of CDCl₃.

8. Accurate volumes were measured and transferred using an Eppendoft® pipette.

9. Symyx HOSS equipment is described on the Symyx website. http://symyx.desantisbreindel.com/page.php?id=71

10. The submitter's equipment and experimental protocol are outlined in this Note and in Notes 11-14. A heavy wall reaction tube (Ace Glass #15 Ace-Tread®, 30 cm length x 38.1 mm O.D., 185 mL capacity) and a 0.5 x 0.125 inch magnetic stir bar are dried in a 125 °C oven overnight. In a glove box, the reaction tube is charged with stock solutions of Rh(acac)(CO)₂ and Bis[(*S*,*S*,*S*)-DiazaPhos-SPE] using a 1000 μ L Eppendoft® pipette followed by 5 grams of substrate. The reaction tube is attached to the reactor head (Note 11). Notably, the addition of the alkene to the catalyst solution resulted in a yellow-white suspension due to partial precipitation of ligand and/or catalyst-ligand complex.

A blast shield must be used whenever the reactor is pressurized and safety procedures for using pressure tubes described in the Ace-Glass® catalog should be reviewed and followed.

The assembled reactor is removed from the glove box, placed in a fume hood, connected to the synthesis gas source and taken through 5 cycles of pressurization (150 psig of 1:1 H₂:CO)/depressurization (0 psig) to replace the nitrogen atmosphere with synthesis gas (Notes 12, 13, Figure 2). The reactor is then submerged in a heated silicon oil bath at the desired temperature. As synthesis gas is consumed, the reactor is repressurized to 150 psi to maintain approximately constant pressure (Note 14). After 2–3 hours (~30–40 psi of synthesis gas consumed) the suspension transforms to a homogeneous yellow solution. In six hours, ~90 psi of synthesis gas is consumed. At the end of the reaction time, the reactor is depressurized.

11. A custom-made reactor head used for hydroformylations is shown in Figure 1. The following parts were used to assemble the reactor head: **a**, Alltech® septum (High-temp, 3/8 in., AT79231) for aliquot-abstractions using a gas-tight syringe, **b**, Swagelok® Brass 1-Piece 40 Series Ball Valve (1.6 Cv, 1/4 in. MNPT x 1/4 in. Swagelok Tube Fitting; product #: B-43M4-

S4), c, Swagelok® Brass Pipe Fitting, Cross (1/4 in. Female NPT; product #: B-4-CS), d, Brass Pipe Fitting, Hex Nipple (1/4 in. Male NPT), e, Swagelok® Brass Pipe Fitting, Elbow (1/4 in. Female NPT; product #: B-4-E), f, Ashcroft® 0-160 psig pressure gauge (1/4 in. NPT, 3.5 in. Dial; McMaster-Carr 3846K311 0-160 psig range), g, Brass Pipe Fitting, Close Nipple (1/4 in. Male NPT), h, #15 Ace-Thred® (15 mm thread, 1/4 in. NPT) PTFE Swagelok adapter; Prod. #: 5844-74), i, Kalrez® 6375 O-ring (9.30) mm x 2.40 mm Part #: K31016K6375), j, #15 Ace Glass® pressure tube (30.5 cm L, 38.1 mm OD, Prod. #: 8648-33), k, Swagelok® Brass 1-Piece 40 Series 3-Way Ball Valve (0.75 Cv, 1/4 in. FNPT; product #: B-43XF4), I, Brass Pipe Fitting (1/4 in. male NPT to 1/4 in. male Swagelok Tube Fitting), m, SS tubing (1/4 in OD, 2 1/2 in. length), and n, Swagelok \mathbb{R} SS Instrumentation Quick-Connect Stem w/ Valve, (0.2 Cv, 1/4 in. Swagelok Tube Fitting, Part #: SS-QC4-D-400). Threads **b**, **d**, **f**, **g**, and **l** were wrapped with PTFE tape prior to assembly. A thorough pressure check of reactor should be taken before conducting an experiment. The most common source of a leak is between the brass pipe fitting \mathbf{g} and the plastic #15 Ace-Thred adapter **h**. Once assembled with the 185-mL pressure tube, the reactor is rather cumbersome to transport—the use of an 11.5" (W) x 13.5" (L) x 5.25" (D) Rubbermaid® dishpan with a 3"(D) x 1" (W) rectangle cut in the tub on the width side was used to partially hold the reactor.

12. A reverse-threaded regulator is connected to a synthesis gas cylinder and Swagelok® Quick-Connects are used to attach to the reactor manifold. The synthesis gas cylinder was obtained from AirGas Inc. as a custom mixture ($48.3\pm2\%$ carbon monoxide balanced with hydrogen gas).

13. The reactor has two possible points of entry: Swagelok® Ball valve **b** fitted with a GC septum, for gas-tight syringe aliquots, and the Swagelok® 3-way Ball Valve **k**, for pressurizing and depressurizing the reactor. In Figure 2, **k** is opened carefully to the synthesis gas cylinder, charging the apparatus to 150 psig (it is advisable to set the regulator on the cylinder to ca. 150 psig and to have a safety shield in place). The valve on **k** is then opened to vent, releasing synthesis gas from the apparatus. After the pressure is reduced to <40 psi, the valve is turned back to the original closed position constituting one cycle. This procedure is repeated for five cycles and the reactor pressure is set at 150 psi. The glass tube of the reactor is lowered into the oil bath for hydroformylation as seen in the far-right picture.

14. Synthesis gas is added manually to maintain at least 100 psig reactor pressure. It is not advisable to maintain reactor pressure by keeping the reactor open to the regulator on the synthesis gas cylinder because, in the event of a leak on the reactor or supply lines, large amounts of H_2 and CO could be released. A carbon monoxide detector is installed near the gas cylinder. Commonly, the synthesis gas line is detached from the reactor at the Swagelok® Quick-Connect during reaction and reconnected when adding more gas. However, if the synthesis gas line is not needed for other reactions, the Swagelok® Quick-Connect system can remain assembled throughout the reaction.

15. ¹H NMR of the crude product mixture indicated >99% conversion of alkene and a branched: linear (2:3) ratio of 2:1.

16. Aldehyde **2** is air-sensitive and flash chromatography should be performed immediately after depressurizing the reactor and the purified product stored in a freezer.

17. Silica gel (250 g) was slurry-packed in a 5-cm diameter column using 5% EtOAc/hexanes. The product was eluted with 5% EtOAc/hexanes, collecting 50 mL fractions, monitored by TLC. (10% EtOAc/hexanes, $R_f 1 = 0.7$, (*R*)-2 = 0.38, 3 = 0.31, visualized with potassium permanganate stain, prepared as follows: 3 g KMnO₄, 20 g potassium carbonate, 5 mL of a 5% (w/w) solution of aqueous sodium hydroxide, and 300 mL of deionized water.) Fractions 9-26 were combined and concentrated by rotary evaporation (40 °C, 20 mmHg) to afford 2 (3.24–3.43 g) as a colorless oil. Fractions 29-38 were combined and concentrated by rotary evaporation (40 °C, 20 mmHg) to afford 3 (1.62–1.70 g) as a colorless oil.

18. (2R)-3-[[(1,1-Dimethylethyl)dimethylsilyl]oxy]-2-methylpropanal **2** has the following physical and spectroscopic data: $[\alpha]_D^{25}$ -34 (c 1.0, CH₂Cl₂); ¹H NMR (500 MHz, CDCl₃) δ : 0.06 (s, -Si(CH₃)₂C(CH₃)₃, 6 H), 0.89 (s, -Si(CH₃)₂C(CH₃)₃, 9 H), 1.10 (d, J = 6.9 Hz, -CHCH₃, 3 H), 2.52– 2.56 (m, -CHCH₃, 1 H), 3.82 (dd, J = 6.4, 10.2 Hz, -CH₂OSi, 1 H), 3.86 (dd, J = 5.2 Hz, 10.2, -CH₂OSi, 1 H), 9.74 (d, J = 1.6 Hz, CHO-CH, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ : -5.33, -5.31, 10.5, 18.4, 26.0, 49.0, 63.7, 204.9; IR (neat): 2957, 2931, 2859, 1736 (C=O), 1473, 1258, 1101, 1033, 838, 778 cm⁻¹; GC-MS *m/z* (relative intensity): 145 (100) [M - *t*-Bu], 115 (95) [SiMe₂*t*-Bu], 101 (31), 85 (25) [Si*t*-Bu], 75 (54), 59 (25); GC purity: 98% (t_R = 7.4 min, same conditions as in note 6); ee 94–96% determined by SFC analysis of benzylamine reductive amination derivative as described in Note 20. The aldehyde oxidizes at a rate of about 1% per week when stored in a - 20 °C freezer.

19. The (S)-enantiomer of **2** was prepared by the same procedure using Bis[(R,R,S)-DiazaPhos-SPE] as ligand; $[\alpha]_D^{25}$ +33 (c 1.0, CH₂Cl₂); ee 88%.

20. The submitters determined chiral purity by gas chromatographic analysis on a Varian Chrompack system using a β -DEX 225 capillary column from Supelco, 30 m x 0.25 mm ID x 0.25 um film thickness. The analytical method used to resolve the enantiomers as follow: 65 °C hold for 70 min, t_R(*R*)-**2**: 60.8 min, t_R(*S*)-**2**: 62.4 min. The checkers determined chiral purity by formation of the reductive amination product with benzylamine and analysis by supercritical fluid chromatography (SFC): tandem columns: 25 cm OZ : 25cm OZ, isocratic 8% 25mM *i*-butylamine in 2-propanol, 100 bar, 2.0 mL/min for 18 min. t_R (*R*)-**4** = 11.5 min, t_R (*S*)-**4** = 14 min.

Procedure for the preparation of the reductive amination product with benzylamine follows.



To a 20-mL vial equipped a 0.7 cm stir bar is added sequentially sodium triacetoxyborohydride (235 mg, 1.1 mmol), chloroform (2 mL), aldehyde 2 (73 mg, 0.36 mmol), and benzylamine (37 mg, 0.34 mmol). The heterogeneous mixture is stirred 16 h at ambient temperature, then quenched with 5 mL sat. NaHCO₃, stirring the biphasic mixture for 5 min. Dichloromethane (10 mL) is added, the layers are separated, and the organic layer dried by filtering through 2 g of sodium sulfate. The filtrate is concentrated by rotary evaporation (40 °C, 20 mmHg) to afford crude 4 (115 mg). The product is purified by silica gel chromatography using 10 g silica with an eluent of 97:2:0.5 CH₂Cl₂:MeOH:Et₃N, collecting 10 mL Fractions 7-10 were combined and concentrated by rotary fractions. evaporation to provide product 4 (68 mg, 67% yield) as a colorless oil having the following physical and spectroscopic data: TLC: $R_f = 0.1$ (97:2:0.5 CH₂Cl₂:MeOH:Et₃N); ¹H NMR (500 MHz, CDCl₃) δ: 0.04 (s, 6 H), 0.89 (s, 9 H), 0.91 (d, J = 6.7 Hz, 3 H), 1.68 (br s, 1 H), 1.85–1.91 (m, 1 H), 2.51 (dd, J = 6.0, 11.6 Hz, 1 H), 2.68 (dd, J = 6.8, 11.6 Hz, 1 H), 3.50-3.57 (m, 2 H), 3.77–3.82 (m, 2 H), 7.25–7.26 (m, 1 H), 7.31–7.33 (m, 4 H); ¹³C NMR (125 MHz, CDCl₃) δ: -5.24, -5.21, 15.6, 18.5, 26.1, 36.1, 53.6, 54.5, 127.0, 128.3, 128.5, 140.9.

21. Linear product **3** has the following spectroscopic data: ¹H NMR (400 MHz, CDCl₃) δ : 0.04 (s, -Si(CH₃)₂C(CH₃)₃, 6 H), 0.88 (s, -Si(CH₃)₂C(CH₃)₃, 9 H), 1.86 (tt, J = 6.0, 7.1 Hz -CH₂CH₂CH₂-, 2 H), 2.50 (dt, J = 7.1, 1.8 Hz, CH₂CHO, 2 H), 3.65 (t, J = 6.0 Hz, -CH₂OSi, 3 H), 9.79 (t, J = 1.7 Hz, CHO, 1 H); ¹³C NMR (100 MHz, CDCl₃) δ : -5.2, 18.5, 25.7, 26.1, 41.0, 62.3, 202.9; GC purity: 96% (t_R = 7.9 min, same conditions as in Note 6)

Safety and Waste Disposal Information

All hazardous materials should be handled and disposed of in accordance with "Prudent Practices in the Laboratory"; National Academies Press; Washington, DC, 2011.

3. Discussion

Protected "Roche Aldehydes" (e.g., 2 (2R)-3-[[(1,1dimethylethyl)dimethylsilyl]oxy]-2-methylpropanal) are common starting materials for the synthesis of polyketides and related molecules.^{3,4} Compared with the common reduction-to-alcohol-followed-by-selective-oxidation-toaldehyde route to 2 from "Roche Ester",⁴ hydroformylation of the protected commodity monomer, allyl alcohol, provides Roche Aldehyde derivatives rapidly, at low cost, and in an easily scalable process. For comparison purposes we have collected the following approximate costs of substrates, normalized to 25 g units, from a common supplier: Roche ester (\$350/25g), allyl alcohol (\$1.00/25g). The only byproducts of the enantioselective hydroformylation of **1** is the corresponding linear aldehyde; although achiral the linear aldehyde is isolated cleanly and constitutes a useful synthetic material also. On larger scales, it should be possible to separate the linear and branched aldehydes by careful vacuum distillation; we have not yet optimized the distillation conditions. An advantage of hydroformylation routes to chiral aldehydes is the absence of acids or bases in the reaction solution that catalyze racemization and condensation reactions. We note that although the Roche Ester has been synthesized by asymmetric hydrogenation of the methyl 2-(hydroxymethyl)-prop-2-enoate,⁵ there is no report of a catalytic hydrogenation route to enantiopure Roche Aldehyde.

Figure 1. The submitters assembled reactor with parts indicated



Figure 2. The submitters' reactor in-use



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- Clark, T. P.; Landis, C. R.; Freed, S. L.; Klosin, K.; Abboud, K. A. J. Am. Chem. Soc. 2005, 127, 5040–5042.
- **3.** For a general review on polyketides stereotetrads, see: Koskinen, A. M. P.; Karisalmi, K. *Chem. Soc. Rev.* **2005**, *34*, 677–690.
- Roche Aldehyde for use in natural product syntheses see: (a) Früstner, A.; Nevado, C.; Waser, M.; Tremblay, M.; Chevrier, C.; Teplý, F.; Aïssa, C.; Moulin, E.; Müller, O. J. Am. Chem. Soc. 2007, 129, 9150– 9161; (b) Canova, S.; Bellosta, V.; Bigot, A.; Mailliet, P.; Mignani, S.; Cossy, J. Org. Lett. 2007, 9, 145–148; (c) Ferrié, L.; Reymond, S.; Capdevielle, P.; Cossy, J. Org. Lett. 2006, 8, 3441–3443; (d) Ehrlich, G.; Kalesse, M. Synlett 2005, 4, 655–657; (e) Smith, A. B., III, Brandt, B. M. Org. Lett. 2001, 3, 1685–1688.
- (a) Qiu, M.; Wang, D.-Y.; Hu, X.-P.; Huang, J.-D.; Yu, S.-B.; Deng, J.; Duan, Z.-C.; Zheng, Z. *Tetrahedron: Asymmetry* 2009, 20, 210–213. (b) Pautigny, C.; Jeulin, S.; Ayad, T.; Zhang, Z.; Genêt, J.-P.; Ratovelomanana-Vidal, V. *Adv. Synth. Catal.* 2008, 350, 2525–2532. (c) Holz, J.; Schäffner, B.; Zayas, O.; Spannenberg, A.; Börner A. *Adv.*

Synth. Catal. **2008**, *350*, 2533–2545. (d) Wassenaar, J.; Kuil, M.; Reek J. N. H. *Adv. Synth. Catal.* **2008**, *350*, 1610–1614. (e) Jeulin, S.; Ayad, T.; Ratovelomanana-Vidal, V.; Genêt, J.-P. *Adv. Synth. Catal.* **2007**, *349*, 1592–1596. (f) Shimizu, H.; Saito, T.; Kumobayashi, H. *Adv. Synth. Catal.* **2003**, *345*, 185–189.

Appendix Chemical Abstracts Nomenclature; (Registry Number)

Silane, (1,1-dimethylethyl)dimethyl(2-propen-1-yloxy)-; (85807-85-8) Silane, chloro(1,1-dimethylethyl)dimethyl-; (18162-48-6)

Prop-2-en-1-ol; (107-18-6)

Rhodium, dicarbonyl(2,4-pentanedionato-κ-O2,κ-O4)-, (SP-4-2)-; (14874-82-9)

Benzamide, 2,2',2",2"'-[1,2-phenylenebis[(1*S*,3*S*)-tetrahydro-5,8-dioxo-1H-[1,2,4]diazaphospholo[1,2-a]pyridazine-2,1,3(3*H*)-triyl]]tetrakis[*N*-[(1*S*)-1-phenylethyl]-; (851770-14-4)

Propanal, 3-[[(1,1-dimethylethyl)dimethylsilyl]oxy]-2-methyl-, (2*R*)-; (97826-89-6),

Butanal, 4-[[(1,1-dimethylethyl)dimethylsilyl]oxy]-; (87184-81-4)



Clark R. Landis was born in Aurora, IL in 1956. After completing his Ph.D. at the University of Chicago in 1983 under the direction of Jack Halpern, Professor Landis held professional positions at the Monsanto Company Corporate Research Lab, the University of Colorado-Boulder, and, since 1990, the University of Wisconsin-Madison. His research interests include bonding theory, computational methods, instrumentation development, chiral ligand synthesis, enantioselective catalysis, catalytic alkene polymerization, and the mechanisms of catalytic reactions.



Gene W. Wong was born in Reno, Nevada in 1985. He received his undergraduate chemistry degree from University of Nevada-Reno, where he conducted research with Prof. Brian J. Frost. He then moved to University of Wisconsin-Madison, where he is currently pursuing a Ph.D. in the research group of Prof. Clark R. Landis as a NSF Predoctoral Fellow. His research focuses on the development of bisdiazaphospholane libraries and rhodium catalyzed hydroformylation.



Tyler T. Adint was born in Fairbanks, Alaska in 1984. He received his undergraduate degree in 2003 from Lewis & Clark College, where he conducted research with Prof. Louis Y. Kuo. He is now pursuing his Ph.D. at the University of Wisconsin-Madison in the research group of Prof. Clark R. Landis. His current research interests concern the synthesis of bisdiazaphospholane ligands that utilize secondary interactions to control selectivity in rhodium catalyzed hydroformylations.

		Peak	2(F1)	[ppm] ?(F1)	[Hz]	Intensity
Comment 1	Data Dawawatawa	1	7.2708	2907.3749	0.18	Inconsicy
	Data Parameters	2	5.9655	2385.4245	0.29	
NAME	32077-152	3	5.9509	2379.5864	0.13	
EXPNO	2 1	4	5.9394	2374.9879	0.33	
PROCNO	1	5	5.9343	2372.9486	0.16	
		6	5.9282	2372.9488	0.18	
1	uisition Parameters	6 7				
Date_	20110409		5.9225	2368.2301	0.28	
Time	13.49	8	5.9113	2363.7516	0.18	
INSTRUM	spect	9	5.9084	2362.5920	0.18	
PROBHD	5 mm QNP 1H/1	10	5.8965	2357.8335	0.26	
PULPROG	zg30	11	5.8852	2353.3150	0.18	
TD	32768	12	5.3057	2121.5903	0.18	
SOLVENT	CDC13	13	5.3007	2119.5910	0.56	
NS	32	14	5.2991	2118.9512	0.23	
DS	2	15	5.2958	2117.6316	0.52	
SWH	6578.947 Hz	16	5.2912	2115.7922	0.16	
FIDRES	0.200774 Hz	17	5.2626	2104.3559	0.15	
AQ	2.4904180 sec	18	5.2579	2102.4765	0.35	
RG	71.8	19	5.2534	2100.6771	0.42	
DW	76.000 usec	20	5.2485	2098.7177	0.12	
DE	7.00 usec	21	5.1118	2044.0555	0.19	
TE	299.6 K	22	5.1072	2042.2161	0.40	
D1	0.10000000 sec	22				
TDO	1		5.1029	2040.4967	0.47	
		24	5.0988	2038.8572	0.17	
	CHANNEL fl =======	25	5.0858	2033.6589	0.18	
NUC1	1H	26	5.0814	2031.8995	0.38	
P1	11.20 usec	27	5.0768	2030.0601	0.45	
PL1	6.00 dB	28	5.0728	2028.4606	0.15	
SF01	399.8724694 MHz	29	4.1987	1678.9342	0.56	
0101	000000000000000000000000000000000000000	30	4.1941	1677.0948	1.23	
F2 - Pro	cessing parameters	31	4.1898	1675.3754	0.66	
SI	16384	32	4.1872	1674.3357	0.86	1
SF	399.8700088 MHz	33	4.1829	1672.6163	1.13	
WDW	no	34	4.1781	1670.6969	0.69	
SSB	0	35	1.5842	633.4741	0.29	
LB	0.00 Hz	36	0.9354	374.0384	1.14	
GB	0.00 112	37	0.9285	371.2793	20.00	
PC	1.00	38	0.9269	370.6395	9.67	1007
PC	1.00	39	0.9209	368.4402		
					1.11	
		40	0.9105	364.0816	0.20	
		41	0.8963	358.4035	0.14	.
		42	0.8783	351.2058	0.24	6.0
		43	0.0941	37.6278	0.53	
		44	0.0869	34.7487	14.04	
		45	0.0854	34.1489	7.16	
		46	0.0790	31.5897	0.69	







32077-152 fr 5-7 nmr400b h-1

Currer NAME EXPNO	nt Data Parameters 32077-152 3									nmr
PROCNC) 1 Acquisition Parameters 20110409 13.56 JM spect 0 5 mm QNP 1H/1 OG zgdc 65536 IT CDC13 835 4 26315.789 Hz			—— I37.77	114.13		77.55	64.32		-0. - Si
 NUC1 PL1 SF01 CPDPRG NUC2 PCPD2 PL2	CHANNEL f1 ====== 13C 4.00 usec 0.00 dB 100.5584512 MHz CHANNEL f2 ======= CHANNEL f2 ======= Waltz16 1H 100.00 usec 120.00 dB									
PL12 SFO2 F2 - F SI SF WDW SSB LB GB PC	24.50 dB 399.8719994 MHz Processing parameters 32768 100.5473690 MHz EM 0 1.00 Hz 0 1.40									
	Peak?(F1)1137.76552114.1298377.5478477.2301576.9122664.3233726.1589818.63039-5.0347	[ppm] ?(F1) 13851.9586 11475.4511 7797.2273 7765.2834 7733.3194 6467.5386 2630.2086 1873.2276 -506.2258	[Hz] 3.40 4.21 5.50 5.70 5.70 5.03 15.00 1.47 7.12	Intensity						
210	200 190 1	80 170 160		130		100 9		70 60	 50	40 30

32077-152 allylOTBS fr 5-7 nmr400b c-13

26.16 18.63 -5.03

Current Data Parameters NAME 32077-169 EXPNO 8 PROCNO 1 F2 - Acquisition Parameters Date_ 20110929 Time 16.54 INSTRUM spect PROBHD 5 mm QNP 1H/13 PULPROG 2g30 TD 65536 SOLVENT CDC13 NS 32 DS 4 SWH 13020.833 Hz FIDRES 0.198682 Hz AQ 2.5166323 sec RG 228.1 DW 38.400 usec DE 6.50 usec TE 300.0 K D1 0.1000000 sec TD0 1 ====== CHANNEL f1 ====== NUC1 1H P1 12.00 usec PL1 -4.00 dB SF01 500.1330885 MHz	Peak 1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33 34	<pre>?(F1) 9.7471 9.7439 7.2704 3.8802 3.8698 3.8597 3.8493 3.8324 3.8197 3.7993 2.5563 2.5528 2.5456 2.5426 2.5320 2.5262 2.5190 2.5156 1.1059 1.0920 1.0868 1.0096 0.9223 0.9039 0.8991 0.8957 0.8824 0.8864 0.8864 0.8804</pre>	[ppm] ?(F1) 4874.8172 4873.2168 3636.1452 1940.6045 1935.4031 1930.3518 1925.1504 1910.3466 1906.4456 1900.1439 1278.4823 1276.7319 1273.1309 1271.5305 1270.1302 1266.3292 1264.9788 1263.4284 1259.8275 1258.1270 553.0938 546.1420 543.5413 504.9313 463.4205 461.2699 452.0675 449.6669 447.9664 446.1660 443.3152 441.4147 440.3145	[Hz] 1.01 1.10 0.62 0.35 0.37 0.76 0.77 0.75 0.30 0.31 0.11 0.09 0.09 0.19 0.11 0.10 0.20 0.10 0.09 0.11 3.19 3.09 0.06 0.15 0.27 0.12 0.08 0.12 0.76 20.00 0.26 0.57
F2 - Processing parameters SI 32768 SF 500.1300083 MHz WDW no SSB 0 LB 0.00 Hz GB 0 PC 1.00	34	0.8804	440.3145	0.57
	35	0.7595	379.8487	0.07
	36	0.1202	60.1156	0.08
	37	0.0674	33.7088	0.20
	38	0.0622	31.1081	5.53
	39	0.0561	28.0573	0.18







Intensity

32077-169 branched fr 16-26 nmr500c h-1

NAME	32077-169
EXPNO	9
PROCNO	1
Date_	20110929
Time	17.08
INSTRUM	spect
PROBHD	5 mm QNP 1H/13
PULPROG	zgdc
TD	131072
SOLVENT	CDC13
NS	992
DS	4
SWH	40322.582 Hz
FIDRES	0.307637 Hz
AQ	1.6253552 sec
RG	8192
DW	12.400 usec
DE	6.50 usec
TE	300.0 K
D1	0.10000000 sec
D1	0.10000000 sec
D11	0.03000000 sec
TD0	40
=======	CHANNEL f1 =======
NUC1	13C
P1	2.50 usec
PL1	0.00 dB
SFO1	125.7703648 MHz
=======	CHANNEL f2 ======
CPDPRG2	waltz16
NUC2	1H
PCPD2	80.00 usec
PL2	120.00 dB
PL12	11.50 dB
SFO2	500.1325007 MHz
SI	65536
SF	125.7577615 MHz
WDW	EM
SSB	0
LB	1.00 Hz
GB	0
PC	1.40

77.48 77.22 76.97

- 63.67

200

.

......

160

150 140 130

120

110

...... 100

.......

60

32077-169 branched fr 16-26 nmr500c c-13



r H H H I I	Current Data Parameters NAME 32077-165 EXPNO 7 PROCNO 1 F2 - Acquisition Parameters Date_ 20110917 Time 8.13	Peak 1 2 3 4 5 6 7 8 9 10 11 12	?(F1) 7.3337 7.3270 7.3236 7.3225 7.3063 7.2705 7.2647 7.2590 7.2580 7.2543 7.2523 7.2472	[ppm] ?(3667.8034 3664.4526 3662.7521 3662.2020 3654.0999 3636.1952 3633.2945 3630.4437 3629.9436 3628.0931 3627.0929 3624.5422						NHCH _O_ _Si- 4	₂Ph ───	
H H H H H H H H H H H H H H H H H H H	INSTRUM spect PROBHD 5 mm QNP 1H/13 PULPROG zg30 ID 65536 SOLVENT CDC13 NS 32 DS 4 SWH 13020.833 FIDRES 0.198682 AQ 2.5166323 SEG 114 DW 38.400 DE 6.50 DE 6.50 DE 6.50 DE 300.0 K 0.10000000 SEC 11 SEC 300.0 NC1 0.10000000 SEC 11 SEC 300.0 FIDO 1 SEC 500.1330885 SFO1 500.1330885 SFO1 500.1330885 SEC 32768	Peak 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33 34 35 36 37	?(F1)) 3.8312 3.8045 3.7936 3.7916 3.7651 3.5602 3.5405 3.5297 3.5171 3.5101 3.4973 2.6980 2.6845 2.6749 2.6614 2.5090 2.4980 2.4859 1.9089 1.8960 1.8834 1.8704	$\begin{array}{c} 1916.0981\\ 1910.6467\\ 1902.7446\\ 1897.2932\\ 1896.2929\\ 1883.0395\\ 1780.5629\\ 1774.9614\\ 1770.7103\\ 1765.3089\\ 1759.0073\\ 1755.5063\\ 1749.1047\\ 1349.3508\\ 1342.5990\\ 1337.7978\\ 1331.0460\\ 1260.9278\\ 1254.8262\\ 1249.3248\\ 1243.2732\\ 954.6982\\ 948.2465\\ 941.9449\\ 935.4432\end{array}$	<pre>.) [Hz] 0.10 0.06 0.04 1.55 1.38 0.06 0.17 0.21 0.66 1.25 0.73 0.18 0.20 0.38 0.39 0.46 0.49 0.41 0.51 0.37 0.41 0.05 0.12 0.20 0.20</pre>	Intensity						,////
2 2 1 0	SF 500.1300082 MHz NDW no SSB 0 LB 0.00 Hz GB 0 PC 1.00	38 39 40 41 42 43 44	1.8580 1.8452 1.6845 0.9174 0.9039 0.8897 0.0452	929.2416 922.8399 842.4690 458.8193 452.0675 444.9657 22.6059	0.12 0.05 0.07 3.04 2.86 20.00 13.29	3.8	3.6 3.4	3.2 3	.0 2.8	2.6 2.	4 2.2	2.0
							0.95	0.90 (0.85 pp	m		
	9.5 9.0 8.5	8.0	1.045 7.5	7.0 6.	5 6.0		5.0 4.	5 4.0	5.145 2.145 3.5	3.0	2.5	2.0

32077-165 (R) chromatograpy fr 4-5 nmr500c h-1



NAME EXPNO PROCNO Date_	32077-165 8 1 20110917										(R) ch fr 4-5 nmr500	1	
Time INSTRUM PROBHD PULPROG TD SOLVENT NS DS	8.22 spect 5 mm QNP 1H/13 zgdc 131072 CDC13 3660 4			140.87	128.52 128.52 128.28			77.48	01.52	54.47			26.13
SWH FIDRES AQ RG DW DE TE D1 D11 TD0	40322.582 H: 0.307637 H: 1.6253552 se 8192 12.400 u: 6.50 u: 300.0 K 0.10000000 se 0.03000000 se 40	z ec sec sec							/		HCH₂P ⊃〔	h	
======= NUC1 P1 PL1 SF01	CHANNEL f1 ====== 13C 2.50 u: 0.00 di 125.7703648 Mi	sec 3								4	I	-	
CPDPRG2 NUC2 PCPD2 PL2 PL12 SFO2 SI SF WDW SSB LB GB PC	CHANNEL f2 ===== waltz16 1H 80.00 u: 120.00 di 11.50 di 500.1325007 Mi 65536 125.7577639 Mi EM 0 1.00 H: 0 1.40	sec 3 1z 1z											
	200 190											·····	

210

200

32077-165 (R) chromatography fr 4-5 nmr500c c-13





Current Data Parameters NAME 32077-169 EXPNO 5 PROCNO 1 F2 - Acquisition Parameters Date_ 20110929 Time 12.41 INSTRUM spect PROBHD 5 mm PABBO BB- PULPROG zg30 TD 32768 SOLVENT CDC13 NS 32 DS 2 SWH 6578.947 Hz FIDRES 0.200774 Hz AQ 2.4904180 sec RG 71.8 DW 76.000 usec DE 6.50 usec TE 300.0 K D1 0.1000000 sec TD0 1 	Peak 1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 34	?(F1) 9.7951 9.7909 9.7865 7.2706 3.6983 3.6834 3.6691 3.6542 3.6393 2.5238 2.5193 2.5017 2.4883 2.4840 2.4667 2.4487 1.8940 1.8790 1.8765 1.8635 1.8616 1.8591 1.8466 1.8591 1.8466 1.8591 1.8466 1.8438 1.8288 0.9079 0.9049 0.8879 0.8889 0.8817 0.0624 0.0524	[ppm] 2(F1) 3921.2724 3919.5910 3917.8296 2910.6393 1480.5405 1474.5755 1468.8508 1462.8859 1456.9210 1010.3529 1008.5514 1003.1870 1001.5056 996.1411 994.4197 987.4940 980.2881 758.2250 752.2201 751.2193 746.0150 745.2543 744.2535 739.2494 738.1285 739.2494 738.1285 732.1235 363.4596 362.2586 359.4563 355.8533 352.9710 24.9806 22.8989	$\begin{array}{c} 0.44\\ 0.92\\ 0.41\\ 0.21\\ 0.10\\ 0.19\\ 0.83\\ 1.46\\ 0.81\\ 0.40\\ 0.38\\ 0.85\\ 0.83\\ 0.41\\ 0.50\\ 0.20\\ 0.10\\ 0.20\\ 0.10\\ 0.23\\ 0.47\\ 0.48\\ 0.29\\ 0.80\\ 0.38\\ 0.44\\ 0.41\\ 0.20\\ 0.38\\ 0.44\\ 0.41\\ 0.20\\ 0.10\\ 0.17\\ 2.36\\ 15.00\\ 1.16\\ 1.71\\ 0.30\\ \end{array}$
SI 16384 SF 400.3300038 MHz WDW no SSB 0 LB 0.00 Hz GB 0 PC 1.00	34	0.0522	20.9773	0.47
	35	0.0447	17.8948	10.14
	36	0.0371	14.8522	0.31







Intensity

32077-169 linear fr 29-38 nmr400b h-1

NAME	32077-169
EXPNO	14
PROCNO	1
Date_	20111125
Time	10.56
INSTRUM &	spect
PROBHD &5	mm PABBO BB-
PULPROG CO	zgdc
TD &	65536
SOLVENT	CDC13
NS &	1308
DS	4
SWH	26315.789 Hz
FIDRES	0.401547 Hz
AQ	1.2452340 sec
RG	8192
DW	19.000 usec
DE	6.50 usec
TE	6.50 usec
D1	673.2 K
D1	0.10000000 sec
D11	0.03000000 sec
TD0	40
NUC1 P1 PL1 PL1W SF01	ANNEL f1 ====== 13C 3.50 usec 0.00 dB 31.90095711 W 100.6741319 MHz ANNEL f2 ====== waltz16 1H 80.00 usec 120.00 dB 17.00 dB 0.00000000 W 0.16438942 W 400.3320017 MHz 32768 100.6630386 MHz EM 0 1.00 Hz 0 1.40

Peak	?(F1)	[ppm]	?(F1)	[Hz]	Intensity	[abs]
1	202.8797		20422.4871	15543	4806.00	
2	77.5507		7806.4891	3035701	06.00	
3	77.2329		7774.4984	3161537	58.00	
4	76.9151		7742.5077	2470566	16.00	
5	62.2663		6267.9150	2619294	36.00	
6	40.9890		4126.0773	2741204	10.00	
7	26.0866		2625.9564	5882066	84.00	
8	25.6786		2584.8859	2650914	24.00	
9	18.4745		1859.6993	9806619	6.00	
10	-5.2200		-525.4611	494051	882.00	

Annotation

210	200	190	180	170	160	150	140	130	120	110	100	90	80	70	60	50	40



-40.99

32077-169 linear nmr400b c-13



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